



Teaching Guide				
Identifying Data				2014/15
Subject (*)	Química Analítica Instrumental 2	Code	610G01014	
Study programme	Grao en Química			
Descriptors				
Cycle	Period	Year	Type	Credits
Graduate	2nd four-month period	Third	Obligatoria	6
Language	Spanish			
Prerequisites				
Department	Química Analítica			
Coordinador	Prada Rodriguez, Dario	E-mail	dario.prada@udc.es	
Lecturers	Andrade Garda, Jose Manuel Prada Rodriguez, Dario Prieto Blanco, Maria del Carmen Rodríguez González, Noelia	E-mail	jose.manuel.andrade@udc.es dario.prada@udc.es m.c.prieto.blanco@udc.es noelia.rodriguez.gonzalez@udc.es	
Web				
General description	<p>Aprendizaxe dos fundamentos básicos, problemas, ventaxas e xestión dalgúns técnicas analíticas instrumentais. Especificamente, abordaranse: técnicas electroanalíticas, técnicas cromatográficas (cromatografía de gases e de líquidos), electroforesis capilar, análise térmico, análises enzimática e inmunoquímica.</p> <p>Aprendizaje de los fundamentos básicos, problemas, ventajas y gestión de algunas técnicas analíticas instrumentales. Especificamente, se abordarán: técnicas electroanalíticas, técnicas cromatográficas (cromatografía de gases y de líquidos), electroforesis capilar, análisis térmico, análisis enzimático e inmunoquímico.</p> <p>The basics, advantages and typical limitations, as well as normal working protocols on several analytical techniques are to be presented. In particular: electroanalytical, chromatographic (gases and liquids), capillar electrophoresis, thermal analysis and enzymatic and immunologic analyses.</p>			

Study programme competences	
Code	Study programme competences
A7	Knowledge and application of analytical methods
A15	Ability to recognise and analyse new problems and develop solution strategies
A16	Ability to source, assess and apply technical bibliographical information and data relating to chemistry
A17	Ability to work safely in a chemistry laboratory (handling of materials, disposal of waste)
A19	Ability to follow standard procedures and handle scientific equipment
A20	Ability to interpret data resulting from laboratory observation and measurement
A21	Understanding of qualitative and quantitative aspects of chemical problems
A22	Ability to plan, design and develop projects and experiments
A23	Critical standards of excellence in experimental technique and analysis
A24	Ability to explain chemical processes and phenomena clearly and simply
A26	Ability to follow standard laboratory procedures in relation to analysis and synthesis of organic and inorganic systems
B1	Learning to learn
B2	Effective problem solving
B3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
B5	Teamwork and collaboration
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life
C6	Ability to assess critically the knowledge, technology and information available for problem solving
C7	Acceptance as a professional and as a citizen of importance of lifelong learning



C8	Understanding role of research, innovation and technology in socio-economic and cultural development
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Learning outcomes			
Subject competencies (Learning outcomes)	Study programme competences		
	<ul style="list-style-type: none"> - Explain adequately the basics and processes related to some fundamental analytical techniques. - Understand their fundamentals, instruments, advantages and limitations. - Get, evaluate and use any source of technical information related to these techniques. - Design and develop strategies to solve analytical problems. - Select the most adequate analytical technique for each particular situation. - Interpret the analytical data. - Get a critical behaviour during the experimental work 	A7	B1
	A15	B2	C3
	A16	B3	C6
	A17	B4	C7
	A19	B5	C8
	A20		
	A21		
	A22		
	A23		
	A24		
	A26		

Contents	
Topic	Sub-topic
Chapter 1: Electroanalytical techniques	Fundamentals of the potentiometric measurements. Fundamentals of polarography and voltamperometry. Electrochemical sensors. Examples Numerical exercises
Chapter 2: Thermal analysis	Fundamentals of thermogravimetry. Fundamentals of differential thermal calorimetry.
Chapter 3: Chromatographic techniques	Fundamentals of gas chromatography. Fundamentals of liquid chromatography. Fundamentals of high resolution liquid chromatography. Examples Numerical exercises
Chapter 4: Electrophoretical techniques	Fundamentals of the electrophoresis Examples
Chapter 5: Enzimatic and immunochemical techniques	Fundamentals of the enzymatic techniques Fundamentals of the immunochemical techniques
Laboratory classes	In total, 20 hours of laboratory classes will be given. They will show the most relevant issues of the instrumentation studied in this subject, taking into account the infrastructure limitations of the Faculty.

Planning			
Methodologies / tests	Ordinary class hours	Student?s personal work hours	Total hours
Laboratory practice	20	22	42
Mixed objective/subjective test	3	0	3
Mixed objective/subjective test	2	0	2
Seminar	7	21	28
Guest lecture / keynote speech	21	52.5	73.5
Personalized attention	1.5	0	1.5

(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.



Methodologies	
Methodologies	Description
Laboratory practice	<p>It is intended that the student works with the analytical techniques studied in the theoretical lessons, taking into account the infrastructure limitations of the Faculty. A laboratory notebook (logbook) has to be kept by the student in order to address his/her practical lessons.</p> <p>The use of leaflets will not be allowed anyway and its use will strongly penalize the final score. A formal notebook has to be used instead.</p>
Mixed objective/subjective test	<p>The objective test to evaluate the knowledge gained by the student will include both theoretical and numerical questions. The former will consist mainly in short questions and one or two questions to be developed longer. They will evaluate the theoretical classes and the seminars. An exam will be made at the end of the first chapters so that (if passed) the student can simplify the first official exam.</p>
Mixed objective/subjective test	<p>Laboratory classes will be evaluated on an ongoing basis. However, by the end of the classes, a test may be delivered to the students in order to evaluate the major issues they obtained. The notebook they kept throughout the classes will also be evaluated and scored. In particular (but not exclusively): order, clarity, correctness in the responses, calculations, graphical schemes, etc.</p>
Seminar	<p>Seminars will be mostly devoted to solve numerical exercises. They must be tried previously by the students so that the seminars would be devoted mainly to solve their doubts.</p>
Guest lecture / keynote speech	<p>The conceptual basis of the different analytical techniques considered in the subject will be reviewed and explained. The underlying chemical, physical or biological bases will be presented. The basic instrumental equipments will be studied and discussed.</p>

Personalized attention	
Methodologies	Description
Laboratory practice Seminar	<p>Students may solve their doubts both during the theoretical classes and the seminars. However, dedicated attention can be obtained in special attention hours (tutorships) which will be agreed between the student and the teacher.</p>

Assessment		
Methodologies	Description	Qualification
Laboratory practice	<p>Daily evaluation: attitude, order, attention, correct experimental work, correct answers.</p> <p>Skills: A7, A15, A16, A17, A19, A20, A21, A22, A23, A24, A26, B1, B2, B3, B4, B5, C1, C3, C6, C7, C8</p>	15
Guest lecture / keynote speech	<p>Attitude and degree of participation of the student in the classes.</p> <p>Skills: A7, A15, A16, A17, A19, A20, A21, A22, A23, A24, A26, B1, B2, B3, B4, B5, C1, C3, C6, C7, C8</p>	5
Seminar	<p>Attitude and degree of participation of the student in the classes. Degree of preparation of the numerical exercises before the seminars.</p> <p>Skills: A7, A15, A16, A20, A21, A22, A23, B1, B2, B3, B4, B5, C1, C3, C6, C7, C8</p>	5
Mixed objective/subjective test	<p>Scoring of the laboratory notebook. Correct answers to the questions. Degree of completeness of the answers and quality of the schemes. In case a final test is performed, score of such a test.</p> <p>Skills: A7, A15, A16, A17, A19, A20, A21, A22, A23, A24, A26, B1, B2, B3, B4, B5, C1, C3, C6, C7, C8</p>	10
Mixed objective/subjective test	<p>Correctness and adequacy in the responses to the theoretical questions.</p> <p>Correct solution to the numerical exercises. Calculations and final exact result.</p> <p>Skills: A7, A15, A16, A20, A21, A22, A23, A24, A26, B1, B2, B3, B4, C1, C3, C6, C7, C8</p>	65

Assessment comments



To pass the subject two basic requisites will be mandatory: (i) regular attendance to all the activities planned for the subject and (ii) get a minimum score on all (and each) activities (5 point out of 10). The objective test will consist of two parts containing theoretical questions and numerical exercises. Each of these two parts are evaluated separately. In any case, the laboratory classes must have been passed in order to pass the subject. After finishing the first chapters, an objective test will be carried out (including theoretical questions and numerical exercises) so that the students passing it (score of 5, out of ten, in each part) may reduce the amount of chapters to be studied for the first examination (first opportunity, May-June). Accordingly, all the scores must yield a minimum sum of 5 (out of 10).

However, note that the subject will not be approved (even when the overall sum exceeds 5) if a particular score does not reach 4. In this case, the final score of the subject will be "fail" (score = 4).

The "Not presented" score will be obtained in case the student makes less than 25% of the academic activities and it does not go to the exam. In any case, to pass the subject the student has to attend, and pass, the laboratory classes.

Note that "continuous evaluation" means that the second opportunity of July is a second opportunity for the exam (Mixed/subjective test). Following, the scores of the laboratory classes, seminars, etc. obtained previously will be maintained. The score of the new exam will substitute that from the first opportunity.

For next courses, no score will be maintained and all activities will have to be repeated.

The maximum score (10, Matricula de Honor) will be obtained by pupils doing the second exam (July) only if that score was not given in the first exam (May-June), according to the Administrative requirements.

Sources of information

Basic	<ul style="list-style-type: none">- RUBINSON, K.A.; RUBINSON, J.J. (2001). Análisis instrumental. Madrid, Prentice Hall- HARRIS, D.C. (2007). Análisis químico cuantitativo. Barcelona, Reverté- KELLNER, R (Editor) (2004). Analytical chemistry. Winheim, Willey- SKOOK, D.A.; WEST, D.M.; HOLLER, F.J. (1996). Fundamentos de química analítica (volumen 2). Barcelona, Reverté- SKOOG, D.A.; HOLLER, F.J.; NIEMAN, T.A. (2001). Principios de análisis instrumental (5a edición). Madrid, McGraw Hill- CHRISTIAN, G.D. (2004). Química analítica (6a edición). México, McGraw Hill
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Complementary

Recommendations

Subjects that it is recommended to have taken before

Química Analítica Avanzada e Quimiometría/610G01015
Medio ambiente e calidade/610G01037

Subjects that are recommended to be taken simultaneously

Subjects that continue the syllabus



Química 1/610G01007

Química 2/610G01008

Química 3/610G01009

Química 4/610G01010

Química Analítica 1/610G01011

Química Analítica 2/610G01012

Química Analítica Instrumental 1/610G01013

Laboratorio de Química/610G01032

Other comments

The subject will not be passed in case the student shows errors in the presentation of the equilibria, estechiometric calculations and / or formulation of chemical equations. For this, the student should be aware of the need of have been studied (and passed) other subjects; at the very least: QA1, QA2, Laboratorio de Química y QAI1

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.