		Teaching	g Guide		
	Identifying	Data			2014/15
Subject (*)	Química Orgánica 1			Code	610G01026
Study programme	Grao en Química				
		Descri	ptors		
Cycle	Period	Yea	ar	Туре	Credits
Graduate	1st four-month period	Seco	ond	Obligatoria	6
Language	SpanishEnglish		'		'
Prerequisites					
Department	Química Fundamental				
Coordinador	Peinador Veira, Carlos E-mail carlos.peinador@udc.es		@udc.es		
Lecturers	García Romero, Marcos Daniel		E-mail	marcos.garcia1@udc.es	
	Peinador Veira, Carlos			carlos.peinador	@udc.es
	Rodriguez Gonzalez, Jaime			jaime.rodriguez	@udc.es
	Ruiz Pita-Romero, Maria			maria.ruiz.pita-r	omero@udc.es
Web	campusvirtual.udc.es/moodle/			1	
General description	A materia pretende proporcionar os co	nnecementos b	pásicos de Química	a Orgánica ao alumno	o do Grao en Química.
	La asignatura pretende proporcionar los conocimientos básicos de Química Orgánica al alumno del Grado en Química.				
	The course provides basics of Organic	Chemistry for	r students of Degre	ee in Chemistry.	

	Study programme competences
Code	Study programme competences
A1	Ability to use chemistry terminology, nomenclature, conventions and units
A4	Knowledge of main types of chemical reaction and characteristics of each
A6	Knowledge of chemical elements and their compounds, synthesis, structure, properties and reactivity
A9	Knowledge of structural characteristics of chemical and stereochemical compounds, and basic methods of structural analysis and research
A10	Knowledge of chemical kinetics, catalysis and reaction mechanisms
A12	Ability to relate macroscopic properties of matter to its microscopic structure
A14	Ability to demonstrate knowledge and understanding of concepts, principles and theories in chemistry
A15	Ability to recognise and analyse new problems and develop solution strategies
A16	Ability to source, assess and apply technical bibliographical information and data relating to chemistry
A21	Understanding of qualitative and quantitative aspects of chemical problems
A24	Ability to explain chemical processes and phenomena clearly and simply
A25	Ability to recognise and analyse link between chemistry and other disciplines, and presence of chemical processes in everyday life
B2	Effective problem solving
В3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
B7	Effective workplace communication
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
C2	Oral and written proficiency in a foreign language
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life
C6	Ability to assess critically the knowledge, technology and information available for problem solving

Learning outcomes	
Subject competencies (Learning outcomes)	Study programme
	competences

Understand basic concepts, principles and theories related to the Organic Chemistry	A1	B2	C1
	A4	В3	C2
	A6	B4	
	A9		
	A10		
	A12		
	A14		
	A15		
	A21		
	A24		
Use and get use to the Organic Chemistry Nomenclature, using main conventions and units	A1	B2	C1
	A6	В3	C2
	A9	В7	C3
	A12		
	A14		
	A24		
	A25		
Understand the caracteristics and main propierties of organic compounds.	A1	B2	C6
	A9	В3	
	A12	B7	
	A14		
	A16		
	A21		
	A24		
Adquire the expertise to solve estructural and synthetic problems in Organic Chemistry being able to relate functional groups	A1	B2	C3
and their reactivities.	A4	В3	C6
	A9	B4	
	A14		
	A15		
	A16		
	A21		
	A24		
	A25		
Understand the most impoortant type of organic reactions, the mechanistic pathways and its stereochemical features.	A1	B2	C1
	A4	B3	C2
	A6	B4	C3
	A9		
	A10		
	A21		
Design and planning in Organic Synthesis. Relationships between spectroscopic data and functional groups in Organic	A1	B2	C3
Chemistry.	A4	B3	C6
	A9	B7	
	A12		
	A14		
	A15		

Adquire expertise in literature searches from Organic Chemistry sources.	A6	В3	C1
	A9	B4	C2
	A14	В7	C3
	A16		
	A24		
	A25		
The Organic Chemistry as a part of our lives	A14	B4	C1
	A15	В7	C2
	A21		C3
	A24		C6
	A25		

	Contents
Topic	Sub-topic
Chemical structure and chemical bonding in organic	Lewis Structures. Resonance Structures. Atomic and hybrid orbitals. Chemical
molecules	structure and bonding on methane, ethene and ethyne.
2. Organic reactions	Characteristics of the organic compounds. Types of reaction Mechanisms. Types of
	reagents: acids and bases, oxidants and reductors, electrophilicity and nucleophilicity.
	Classification of the Organic reactions. Thermodynamic and kinetic features of
	Organic Reactions. Kinetic and thermodynamic control. Reaction intermediates.
3. Alkanes and cycloalkanes. Reactions	Physical properties. n-Alkanes: Rotational barriers. Conformational isomerism.
	Cycloakanes: ring strain.
4. Stereochemistry	Optical activity. Chirality and enantiomers. Nomenclature. Diasteroisomers.
	Stereoisomerism in cyclic molecules.
5. NMR Spectroscopy	Basic principles of the NMR. The most important nuclei studied in Organic Chemistry.
	The chemical shift, Spin-spin coupling: N+1 rule. Identification of Organic functional
	groups by NMR.
6. Reactions of Alkanes.	Halogenation. Pyrolisis. Combustion. Cracking.
7. Alkyl Halides	Chemical structure and properties. Nucleophilic substition reactions (SN). Factors
	determining the SN mechanisms: substrate (structure of the alkyl group and nature of
	the leaving group, nucleophilicity of the reagent and solvent effects. Competitive
	proccesses in the SN reaction: transpositions and eliminations. Organometallic
	compounds. Reduction of alkyl halides.
8. Alcohols	Chemical structure. Acid-base propierties. Reactions through the O-H bond. Reactions
	through the C-O bond. Oxidation. Thiols.
9. Ethers	Chemical structure. Ether cleavage. Epoxides. Thioethers.
10. Amines	Structure and properties. Acid-base reactions. Alkylation of amines. Hoffman's
	elimination. Oxidation: Cope's elimination.

Planni	ng		
Methodologies / tests	Ordinary class	Student?s personal	Total hours
	hours	work hours	
Mixed objective/subjective test	4	0	4
Seminar	10	25	35
Guest lecture / keynote speech	20	30	50
ICT practicals	10	10	20
Problem solving	10	30	40
Personalized attention	1	0	1
(*)The information in the planning table is for guidance only and does no	t take into account the	heterogeneity of the stu	idents.

	Methodologies
Methodologies	Description
Mixed	A final exam will be introduced at the end of the semester. This is planned to objectively asses the degree of understanding by
objective/subjective	the student, as well as the ability to apply the course contents. This test will include a single type of questions (related to the
test	structure, structural elucidation, reactivity and synthesis of organic compounds), in order to determine whether the answers are
	correct or not.
Seminar	Sessions devoted to the resolution of problems and questions related to the course contents, with the active participation of
	students. This methodology is intended to the whole group.
Guest lecture /	The teacher presents the basic contents of each unit. These materials will be provided in advance to the students in order to
keynote speech	study them before the development of the class. This methodology is intended to the whole group.
ICT practicals	Practicals will be mainly focused on two aspects:
	1. The use of software tools for the three-dimensional representation of organic compounds, and the application of such tools
	on solving stereochemistry and conformational analysis problems.
	2. The structural elucidation of organic compounds by means of proton NMR with the support of software for the simulation of
	spectra.
Problem solving	The teacher will assign students the preparation of some problems, that would require the integration of contents from different
	subjects. Then, students should study the problems and prepare a solution in writing, which must be submitted to the teacher
	prior to the development of this activities in the classroom. During the classes, students will present orally some of the
	problems assigned, and will have to answer to issues that may arise in this regard.

	Personalized attention
Methodologies	Description
Seminar	Personal attention for each student will be integrated on this methodology during interviews before laboratory practices.
ICT practicals	
Problem solving	

	Assessment	
Methodologies	Description	Qualification
Mixed	Final written test. Competencies assessed: A1, A6, A9, A12, A14, B2, B3	70
objective/subjective		
test		
Seminar	The evaluation will consider the attendance, participation, and completion of a colection of practice tests of	5
	proposed problems. Competencies assessed: A1, A4, A10, A15	
ICT practicals	A necessary requirement for passing the entire course is to achieve a passing score in the ICT practicals.	10
	Regular attendance to all of your sections is a must. The attendance, follow up and participation on the	
	practical sessions will contribute a 5% to the final grading.	
	In the last training session within the practicals, the students will solve a similar problem to those presented,	
	contributing another 5% to the evaluation. Competencies assessed: A1, A9, A16, B2, B3, B4, C1, C2, C3, C6.	
Problem solving	Attendance, participation, and the quality of the written solutions in the take home assigned problems will be	15
	evaluated. The students will deliver the written solutions before of each presentation. Additionally the clarity	
	and precision of explanations, as well as the use of appropriate nomenclature for the compounds and	
	reactions it will be evaluated.	
	Competencies assessed: A1, A4, A6, A9, A10, A21, A24, A25, B2, B3, B4, B7.	

Assessment comments

The assistance to the ICT practical is a necessary requirement for passing the course.

subjects of each topic as preparation to the teacher's keynote lectures.

The final grade will be based on your performances in the all activities. The exam will make up 70%, seminar 5%, ICT practicals 10%, and problem solving 15%. Each activity (except for seminar) should reach a minimum score of 4 out of 10. A final score of 5 out of 10 is needed in order to pass the entire course.

The scores obtained on the practicals, seminars and problem-solving classes will be preserved to the "second opportunity" of an academic course. Those students not participating in more than a 25% of the planned activities will obtain an assesment of "not attended".

Students evaluated in the "second chance" shall be eligible for "class honors" if the maximum number of those marks for the corresponding course has not been fully covered in the "first opportunity".

Regarding the successive academic years, the teaching-learning process (including the assessment), it refers to an academic course, and therefore would re-start with a new course, including all the planified activities and evaluation procedures.

	Sources of information
Basic	- E. QUIÑOÁ y R. RIGUERA (2004). CUESTIONES Y EJERCICIOS DE QUÍMICA ORGÁNICA (2ª ed). Madrid,
	McGraw-Hill
	- K.P.C. VOLLHARDT and N.E. SCHORE (2007). Organic Chemistry (5ª ed). Barcelona, Omega
	- L.G. WADE, Jr. (2004). QUÍMICA ORGÁNICA (5ª ed). Madrid, Pearson Educación
Complementary	

Recommendations
Subjects that it is recommended to have taken before
Química Orgánica 2/610G01027
Ampliación de Química Orgánica/610G01028
Experimentación en Química Orgánica/610G01029
Química Orgánica Avanzada/610G01030
Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus
Química 1/610G01007
Química 2/610G01008
Química 3/610G01009
Química 4/610G01010
Other comments
We highly recommend as main sources of information the books: Wade 2004 and Vollhardt, 2007. We also recommend the reading of the main

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.