

			Teachin	g Guide			
		ldentifyir	ng Data			2014/15	
Subject (*)	Ciencia de Materiais Code			610G01035			
Study programme	Grao en	Química					
			Desci	riptors			
Cycle		Period	Ye	ear	Type	Credits	
Graduate		1st four-month period	Fou	urth	Obligatoria	6	
Language	Spanish	GalicianEnglish					
Prerequisites							
Department	Química	Fundamental					
Coordinador	Castro G	Sarcia, Socorro		E-mail	socorro.castro.ga	arcia@udc.es	
Lecturers	Castro G	Sarcia, Socorro		E-mail	socorro.castro.ga	arcia@udc.es	
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Web							
General description	- Galego	o: Ciencia de Materiais é ur	nha materia obr	igatoria do primeir	o semestre do carto cur	so do Grao en Química.	
	Dita mat	eria pertence ó módulo de F	ormación com	plementaria en Qu	ímica. Esta materia, de	carácter claramente	
	interdisc	iplinar, está adicada ó estud	lo dos distintos	tipos de materiais	(metálicos, cerámicos,	poliméricos, materiais	
	compost	os), as súas propiedades (	mecánicas, elé	ctricas, magnéticas	s, ópticas, etc.) e as súa	as prestacións.	
	- Espar	ñol: Ciencia de Materiales e	s una materia d	obligatoria del prim	er semestre del cuarto	curso del Grado en Química.	
	Dicha as	signatura pertenece al módu	lo de Formació	n complementaria	en Química. Esta asign	atura, de carácter claramente	
	interdisc	iplinar está dedicada al estu	ıdio de los distir	ntos tipos de matei	riales (metálicos, cerám	icos, poliméricos, composites),	
	sus prop	iedades (mecánicas, eléctri	cas, magnética	s, ópticas, etc.) y s	sus prestaciones.		
	- English	: Materials Science is a con	npulsory subjec	et of the 7th semes	ter (4th year, 1st semes	ter) of our Degree in Chemistry,	
	that belo	ongs to the module "Co	omplementary F	Formation in Chem	istry". This subjec	t, of marked interdisciplinary	
	characte	er, is devoted to the study of	the different typ	pes of materials (m	netals, ceramics, polyme	ers, composites), their properties	
	(mechan	nical, electrical, magnetic, op	tical) and their	performance.			

	Study programme competences
	Study programme competences
Code	Study programme competences
A1	Ability to use chemistry terminology, nomenclature, conventions and units
А3	Knowledge of characteristics of the different states of matter and theories used to describe them
A4	Knowledge of main types of chemical reaction and characteristics of each
A5	Understanding of principles of thermodynamics and its applications in chemistry
A6	Knowledge of chemical elements and their compounds, synthesis, structure, properties and reactivity
A7	Knowledge and application of analytical methods
A9	Knowledge of structural characteristics of chemical and stereochemical compounds, and basic methods of structural analysis and
	research
A10	Knowledge of chemical kinetics, catalysis and reaction mechanisms
A12	Ability to relate macroscopic properties of matter to its microscopic structure
A14	Ability to demonstrate knowledge and understanding of concepts, principles and theories in chemistry
A15	Ability to recognise and analyse new problems and develop solution strategies
A16	Ability to source, assess and apply technical bibliographical information and data relating to chemistry
A17	Ability to work safely in a chemistry laboratory (handling of materials, disposal of waste)
A18	Risk management in relation to use of chemical substances and laboratory procedures
A19	Ability to follow standard procedures and handle scientific equipment
A20	Ability to interpret data resulting from laboratory observation and measurement
A21	Understanding of qualitative and quantitative aspects of chemical problems

A22	Ability to plan, design and develop projects and experiments
A23	Critical standards of excellence in experimental technique and analysis
A24	Ability to explain chemical processes and phenomena clearly and simply
A25	Ability to recognise and analyse link between chemistry and other disciplines, and presence of chemical processes in everyday life
A26	Ability to follow standard laboratory procedures in relation to analysis and synthesis of organic and inorganic systems
A28	Acquisition, assessment and application of basic principles of industrial activity, organisation and task management
B1	Learning to learn
B2	Effective problem solving
В3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
B5	Teamwork and collaboration
В6	Ethical, responsible, civic-minded professionalism
B7	Effective workplace communication
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life
C4	Self-development as an open, educated, critical, engaged, democratic, socially responsible citizen, equipped to analyse reality, diagnose
	problems, and formulate and implement informed solutions for the common good
C5	Understanding importance of entrepreneurship, and knowledge of resources available for people with business ideas
C6	Ability to assess critically the knowledge, technology and information available for problem solving
C7	Acceptance as a professional and as a citizen of importance of lifelong learning
C8	Understanding role of research, innovation and technology in socio-economic and cultural development

Learning outcomes			
Subject competencies (Learning outcomes)	Study	/ progra	amme
	cor	npeten	ces
To achieve a general vision of Materials Science and its interdisciplinary character . To achieve basic notions about different	A15	B1	C1
criteria for the classification of materials.	A25	B2	C3
		В3	C4
			C5
			C6
			C7
			C8
To know the different families of materials (metals, ceramics, polymers, composites) and the main methods of synthesis and	A1	B1	C1
processing	A5	B2	C3
To know their most outstanding properties and to be able to relate them to their composition, structure and microstructure.	A9	В3	C4
	A10		C5
	A12		C6
	A15		C7
	A16		C8
	A21		
	A24		
	A25		

To know the basics and applications of materials of interest for the electrical and electronic industries, and also of magnetic	A1	B1	C1
and optical materials.	А3	B2	C3
	A4	В6	C4
	A5		C5
	A6		C6
	A9		C7
	A12		C8
	A14		
	A15		
	A16		
	A21		
	A24		
	A25		
To develop criteria for the selection of materials based on their applications.	A14	B1	C1
	A15	B2	C3
	A21	В3	C4
	A24	B4	C5
	A25	B5	C6
	A28	В6	C7
		B7	C8
To get acquainted with techniques and working methodologies of common use in Materials Labs.	A1	B1	C1
	A6	B2	C3
	A7	В3	C4
	A9	B4	C5
	A10	B5	C6
	A12	B6	C7
	A16	B7	C8
	A17		
	A18		
	A19		
	A20		
	A21		
	A22		
	A23		
	A24		
	A25		
To be able to apply a and interpret date containing ecceptific and technical information about materials	A26	D4	C4
To be able to analyze and interpret data containing scientific and technical information about materials.	A1	B1	C1
	A15	B2	C3
	A16	B3 B4	C4 C5
		B6	C6

Contents		
Topic	Sub-topic	
I. Introductory Section	? Introduction to Materials Science	
	? Criteria for classification of Materials	
	? Criteria for selection of Materials	
	? Mechanical tests and properties	
	? Structure, microstructure and phase diagrams	

II. Families of Materials	? Metals and alloys (steels, cast irons and non-ferrous alloys)
	? Ceramics (clay products, refractories, abrasives, cements, glasses, advanced
	ceramics)
	? Polymers (thermoplastics, termosetting polymers, elastomers)
	? Composites
	?Synthesis and processing tecniques
III: Materials of technological interest	? Materials for the electrical and electronic industries: metals, semiconductors,
	superconductors, dielectrics, ferroelectrics, piezoelectrics, ionic conductors, etc.
	Devices.
	? Magnetic materials: ferromagnetic, ferrimagnetic and antiferro-magnetic.Hard and
	soft magnetic materials. Main apllications (motors, magnetic storage, etc.)
	? Introduction to optical materials and their applications
IV: New trends in Materials Science and examples of selection	? Biomaterials, nanomaterials, etc.
of materials	? Examples of selection of materials
V: Materials Laboratoy	? Synnthesis and processing
	? Characterization and study of materials properties

	Planning		
Methodologies / tests	Ordinary class	Student?s personal	Total hours
	hours	work hours	
Guest lecture / keynote speech	25	62.5	87.5
Problem solving	5	12.5	17.5
Seminar	2	6	8
Events academic / information	1	2	3
Supervised projects	2	5	7
Laboratory practice	9	9	18
Mixed objective/subjective test	3	4.5	7.5
Personalized attention	1.5	0	1.5

	Methodologies
Methodologies	Description
Guest lecture /	In these lectures the teacher will present the contents of the different themes, enphasizing their main aspects and paying
keynote speech	special attention to fundamental and/or difficult concepts.
Problem solving	These classes will be devoted to the resolution of a series of problems and questions that will be given in advance to the
	student so that he/she can work on them before the corresponding session.
Seminar	Work in small groups devoted to the discussion of a topic, a case study etc. prepared in advanced by the students.
Events academic /	Complementary activities consisting in a visit to centers/installations working in aspects related to this subject, attendance to
information	scientific talks organized by the Faculty and other organisms during this semester, etc.
	These activities will be organized and announced during the course depending on the number of students, the invited
	speakers that will come tro the Faculty during this period, etc.
Supervised projects	Before starting the work in the Lab the students will have to carry out a study about aspects of interest for the topic of his/her
	experiment. This task will be guided and supervised by the teacher with whom the student will have at least one personal
	interview.
	Also, in the case of preparation of a case study or similar activity the students will have at least one tutoring session with the
	teacher that will orientate and supervise their work in prorgress
Laboratory practice	Supervised work that will be carried out in the Lab and that will be centered in the synthesis and processing of different types
	of materials, their characterization and study of their properties.



Mixed	Final exam that will take place on the official dates approved by the "Junta de Facultad". It will evaluate the level of
objective/subjective	knowledge and competences acquired by the student ands his/her capability to integrate them as well as proof a global vision
test	on the subject.

	Personalized attention
Methodologies	Description
Guest lecture /	The personalized attention to students, understood as a support in the teaching-learning process, will take place in the
keynote speech	sessions carried out with small groups and individual interviews. In addition, the student can ask for additional tutoring
Seminar	sessions that will take place during the tutoring hours of the teacher (whose timetable will be indicated at the beginning of the
Events academic /	course)
information	
Laboratory practice	
Problem solving	
Supervised projects	
Mixed	
objective/subjective	
test	

	Assessment	
Methodologies	Description	Qualification
Guest lecture /	Aspects that will be taken into account: extent of previous preparation, of follow-up and of active participation	0
keynote speech	A:1,3,4,5,6,7,9,10,12,14,24,25,27	
	B:1,3,6,7	
	C:1,6,8	
Seminar	Aspects that will be taken into account: the work done by the students, their answers, their level of knowledge	0
	and their active participation in debates with the other students .	
	A:1,3,4,5,6,7,9,10,12,14,15,21,24,25	
	B:1,2,3,4,5,7	
	C:1,3,4,6,8	
Events academic /	Aspects that will be taken into account: the conclusions that the students have extracted from these activities	0
information	and that they should also refect in a written summary that they will have to present afterwards.	
	A:25,28	
	B:1,3	
	C:1,4,5,6,7,9	
Laboratory practice	The work done in the lab will be assessed from the point of view of: organization and security, knowledge of	20
	material and techniques, working skills, and specially degree of understanding and rationalization of observed	
	facts on a scientific basis. The previous preparation of this work and the laboratory notebook will be also taken	
	into account in this evaluation.	
	Although the grade will be based on a continuous evaluation, in particular cases the teacher can also carry out	
	an exam.	
	A:1,3,4,5,6,7,9,10,12,14,15,16,17,18,19,20,22,24,25,26,27	
	B:1,2,3,4,5,6,7	
	C:1,4,6,8	
	"Laboratory practice" and "supervised projects" will be assessed together.	

Problem solving	Aspects that will be taken into account: the response of the students and their active participation in presencial	20
	activities. Occasionally, and if requested by the teacher, the students should submit their answers to selected	
	questions, that can also be evaluated.	
	A:1,3,4,5,6,7,9,10,12,14,15,21,24,25	
	B:1,2,3,4,5,7	
	C:1,3,4,6,8	
	"Problem solving", "seminars" and "events" will be assessed together.	
Supervised projects	In the tutoring sessions associated to the activity "Trabajos tutelados" the teacher will not only	0
	guide the students but will also evaluate all aspects related to their preparation for the work in the Lab (in fact	
	the student will not be allowed to start the work in the Lab until he/she have done a goodenough previous	
	preparation), for the case studies , etc.	
	A:1,3,4,5,6,7,9,10,12,14,16,20,22,23,24,25,27	
	B:1,2,3,4,7	
	C:1,3,4,6,8	
Mixed	This final exam, that will take place at the end of the semester, can contain: questions to develop, short	60
objective/subjective	questions, a test and problems that will be similar to those solved during the Course.	
test	A:1,3,4,5,6,7,9,10,12,14,24,25	
	B:1,2,3,4,6	
	C:1,6	

## **Assessment comments**

The final grade will result from the addition of the following partial contributions:

- Final exam ("Prueba mixta"): up to a maximun of 6 points.
- -Activities carried out during the different sessions (seminars, exercises, tutoring sessions, scientif events, etc): up to a maximum of 2 points.
- -Laboratory sessions: up to a maximum of 2 points.

To pass this subject aminimum of 5 points will be required, with the restriction that in the final exam ("prueba mixta") aminimum of 2.4 (over a maximum of 6) will be necessary as well as a minimum of 0.8 (over a maximum of 2) in the laboratory sessions. If these minima are not achieved the students will fail

As the assessment of this subject is based on a continuous evaluation model, the progression of the student throughout the semester will be taken into consideration with a maximum of 1 point.

Also, and according to this evaluation model, if the student has participated in activities whose relativeweight is more than a 25% of the total grade, he/she will be assessed. In this same context, and according to the rules contained in ?Probas de Avaliación eActas de Cualificación de Grao e Mestrado?, the so-called ?second opportunityof July? is understood as a second opportunity to carry out a final exam ("prueba mixta"). Nevertheless, and in very special cases, theteacher could also include a second part concerning aspects of the Laboratory Sessions.

This mark will be considered together with the others obtained during the course corresponding to the other activities (seminars, exercises, scientific events, etc.). The percentages of the different contributions will be the same as those of the former "first opportunity".

The highest grade "Matricula de Honor" will be mainly given to students that pass the subject in the First Opportunity". And it will only be given in the so-called "second Opportunity" if there are still any available. In the case of very special and exceptional circumstances, adequately justified, the teacher can totally or partially exempts the student from the continuous evaluation process. This student will then have to go through an examination process where he/she will need to clearly proof his/herlevel of knowledge, competence, capabilities and skills.

	Sources of information
Basic	
Complementary	

Recommendations
Subjects that it is recommended to have taken before



	Subjects that are recommended to be taken simultaneously
	Subjects that are recommended to be taken simultaneously
	Subjects that continue the syllabus
Química Inorgánica 1/610G01021	
Química Inorgánica 2/610G01022	
Química Inorgánica 3/610G01023	
Química Inorgánica 4/610G01024	
	Other comments

(\*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.