

		Teaching Guide				
	Identifying I	Data		2015/16		
Subject (*)	Propiedades de Materiais		Code	610509020		
Study programme	Mestrado en Investigación Química	Mestrado en Investigación Química e Química Industrial				
	-	Descriptors				
Cycle	Period	Year	Туре	Credits		
Official Master's Degre	ee 1st four-month period	First	Optativa	a 3		
Language	Galician			I		
Teaching method	Face-to-face					
Prerequisites						
Department	Química Fundamental					
Coordinador		E-i	nail			
Lecturers	Sanchez Andujar, Manuel	E-i	nail m.anduja	r@udc.es		
Web		I	I			
General description	This subject is important in the modu	le Nanochemistry and	New materials, where i	t will reported the theoretical foundation		
	of many of the fundamental propertie	es of materials, which w	ill be studied in other s	ubjects of this module.		

	Study programme competences / results
Code	Study programme competences / results
A1	Define concepts, principles, theories and specialized facts of different areas of chemistry.
A3	Apply materials and biomolecules in innovative fields of industry and chemical engineering.
A9	Promote innovation and entrepreneurship in the chemical industry and in research.
B1	Possess knowledge and understanding to provide a basis or opportunity for originality in developing and / or applying ideas, often within a
	research context
B4	Students should be able to communicate their conclusions, and the knowledge and the reasons that support them to specialists and
	non-specialists in a clear and unambiguous manner
B5	Students must possess learning skills to allow them to continue studying in a way that will have to be largely self-directed or autonomous.
B7	Identify information from scientific literature by using appropriate channels and integrate such information to raise and contextualize a
	research topic
B10	Use of scientific terminology in English to explain the experimental results in the context of the chemical profession
B11	Apply correctly the new technologies to gather and organize the information to solve problems in the professional activity.
B13	Assess the human, economic, legal and ethical dimension in professional practice as well as the environmental implications of their work

Learning outcomes			
Learning outcomes	Study	y progra	mme
	con	npetence	es/
		results	
Understanding the fundamental aspects of the theory of solid, in relation to electronic structure and lattice.	AC1	BC1	
	AC3	BC4	
	AC9	BC7	
		BC10	
		BC11	
Use the existing relations between the fundamental aspects of the theory and the different electronic properties and network	AC1	BC1	
with the experimental findings.	AC3	BC4	
	AC9	BC5	
		BC10	
		BC11	



Understanding the influence of the c	dimensionality of the system on these properties.

AC1	BC7	
AC3	BC10	
AC9	BC11	
	BC13	

Contents			
Торіс	Sub-topic		
Topic 1 Classic and quantum models of free electrons.	Drude model and Sommerfeld model		
	O modelo de Drude e o modelo de Sommerfeld		
	Effect of periodic potential in the properties of the electron gas		
Topic 2 Quantization of lattice vibrations	Phonons		
Topic 3 Experimental techniques to determining the	Electrical conductivity, thermal conductivity, thermoelectric power and Hall effect.		
properties of electrical and thermal transport.			
Topic 4Cooperative phenomena in insulatings.	Ferroelectricity and localised magnetism.		
Tema 5 Optical properties of matter	General trends. Optical properties of metals and semiconductors.		

Plannin	g		
Competencies /	Teaching hours	Student?s personal	Total hours
Results	(in-person & virtual)	work hours	
A1 A3 A9 B1 B4 B5	4	12	16
B7 B10 B11 B13			
A1 A3 A9 B1 B4 B5	1	6	7
B7 B10 B11 B13			
A1 A3 A9 B1 B4 B5	1	6	7
B7 B10 B11 B13			
A1 A3 A9	15	30	45
	0	0	0
	Competencies / Results   A1 A3 A9 B1 B4 B5   B7 B10 B11 B13   A1 A3 A9 B1 B4 B5   B7 B10 B11 B13   A1 A3 A9 B1 B4 B5   B7 B10 B11 B13   A1 A3 A9 B1 B4 B5   B7 B10 B11 B13   A1 A3 A9 B1 B4 B5   B7 B10 B11 B13	Results (in-person & virtual)   A1 A3 A9 B1 B4 B5 4   B7 B10 B11 B13 1   A1 A3 A9 B1 B4 B5 1   B7 B10 B11 B13 1   B7 B10 B11 B13 1   B7 B10 B11 B13 1   A1 A3 A9 B1 B4 B5 1   B7 B10 B11 B13 1   A1 A3 A9 B1 B4 B5 1   B7 B10 B11 B13 1   A1 A3 A9 15	Competencies / ResultsTeaching hours (in-person & virtual)Student?s personal work hoursA1 A3 A9 B1 B4 B5412B7 B10 B11 B1316B7 B10 B11 B1316B7 B10 B11 B1316B7 B10 B11 B1316B7 B10 B11 B1316A1 A3 A9 B1 B4 B516B7 B10 B11 B133015

(\*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

	Methodologies
Methodologies	Description
Seminar	Practical lessons in which it proposes and solves problems, exercises, etc. The student participates actively in different ways:
	delivery of exercises to the teacher, exhibition of work, practical classes, etc. Attendance to these lessons are mandatory.
Supervised projects	The student will perform the exposure of previously proposed work. The schedule will be agreed with the students.
Mixed	Final test to help assess the level of knowledge and skills acquired by students.
objective/subjective	
test	
Guest lecture /	Lesson taught by the teacher who may have different formats (theory, problems and / or general examples, general guidelines
keynote speech	on the matter). The teacher may have the support of audiovisual and computer media.

	Personalized attention
Methodologies	Description
Seminar	All students must participate in an active way in these activities, so that teachers can check if the students are acquiring the
Supervised projects	skills of this subject.

		Assessment	
Methodologies	Competencies /	Description	Qualification
	Results		



Mixed	A1 A3 A9 B1 B4 B5	Exam or objective test	60
objective/subjective	B7 B10 B11 B13		
test			
Guest lecture /	A1 A3 A9	Active participation during the guest lectures.	2
keynote speech			
Seminar	A1 A3 A9 B1 B4 B5	Solving the proposed problems.	30
	B7 B10 B11 B13		
Supervised projects	A1 A3 A9 B1 B4 B5	Resolution and / or presentation of the supervised works.	8
	B7 B10 B11 B13		

Assessment comments

The evaluation of the subject is by continuous assessment and a final test. Continuous assessment has a weight of between 25% and 40% in the qualification. The rest will sign the result of the final test.

	Sources of information	
Basic - S. Elliot (). The Physics and Chemistry of Solids.		
	- P. A. Cox (). The Electronic Structure and Chemistry of Solids.	
	- J. M. Ziman (). Principles of the Theory of Solids.	
Complementary	- J. B. Goodenough (). Magnetism and the Chemical Bond.	
	- C. F. Bohren and D. R. Huffman (). Absorption and Scattering of light by small particles.	

Recommendations
Subjects that it is recommended to have taken before
Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus
Other comments
s very important to attend all classes Problem solving and self-assessment exercises is key in learning this subject. It may be useful to start wit

- It is very important to attend all classes.- Problem solving and self-assessment exercises is key in learning this subject. It may be useful to start with problems resolved in the manuals of support and referral to follow later with problems proposed at the end of each chapter in the reference manuals.- It is essential to consult the bibliography and try to complete with advanced aspects most fundamental concepts that are explained in class.

(\*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.