



Teaching Guide				
Identifying Data				2015/16
Subject (*)	Enxeñaría Química	Code	610G01033	
Study programme	Grao en Química			
Descriptors				
Cycle	Period	Year	Type	Credits
Graduate	1st four-month period	Third	Obligatoria	6
Language	Spanish			
Teaching method	Face-to-face			
Prerequisites				
Department	Química Física e Enxeñaría Química 1			
Coordinador	Kennes , Christian	E-mail	c.kennes@udc.es	
Lecturers	Kennes , Christian Vega Martin, Alberto de	E-mail	c.kennes@udc.es alberto.de.vega@udc.es	
Web				
General description	La asignatura describe los conceptos básicos de la Ingeniería Química (operaciones unitarias, balances de materia, energía y cantidad de movimiento, fundamentos de fenómenos de transporte, y reactores químicos)			

Study programme competences	
Code	Study programme competences
A11	Knowledge and design of unit operations in chemical engineering
A15	Ability to recognise and analyse new problems and develop solution strategies
A19	Ability to follow standard procedures and handle scientific equipment
A20	Ability to interpret data resulting from laboratory observation and measurement
A25	Ability to recognise and analyse link between chemistry and other disciplines, and presence of chemical processes in everyday life
B2	Effective problem solving
B5	Teamwork and collaboration
C2	Oral and written proficiency in a foreign language
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life

Learning outcomes			
Learning outcomes	Study programme competences		
Know the fundamentals of unit operations in Chemical Engineering and of their design	A11 A15 A19 A20 A25	B2 B5	C2 C3
Apply mass and energy balances to unit operations and (bio)reactors	A11 A15 A19 A20 A25	B2 B5	C2 C3
Know the fundamentals of applied kinetics and of the design of (bio)reactors	A15 A19 A20	B2 B5	C2 C3



Know the fundamentals of mass transfer and heat transfer	A11	B2	C2
	A15	B5	C3
	A19		
	A20		
	A25		

Contents	
Topic	Sub-topic
1. Introduction to Chemical Engineering.	Fundamentals of chemical engineering. Representative examples of processes in the chemical industry. Definitions of common use: (non) continuous operation, (non) steady-state, equilibrium stages, contact between phases, etc.
2. Fundamentals of unit operations.	Classification of unit operations. Mass transfer-, heat transfer-, simultaneous mass and heat transfer-, and momentum transfer- operations. Representative examples of unit operations. Equipment description.
3. Transport phenomena.	Mass transport. Heat transfer. Momentum transfer. Fundamentals of rheology. Viscosity. Analogy between different transfer processes and their governing laws. Examples.
4. Introduction to balance equations.	General problem-solving strategies. Different types of balances. Dimensions, units, and their conversion.
5. Mass balances on non-reactive processes.	General case. Recycle, purge, and by-pass. Steady- and non-steady- state.
6. Mass balances on reactive processes.	Simple and multiple reactions. Recycle, purge, and by-pass. Steady- and non-steady-state..
7. Energy balances.	Forms of energy. Fundamentals of energy balances. Steady- and non-steady- states.
8. Chemical reactors and bioreactors.	Ideal batch reactors and continuous reactors. Constant and variable volume/density reactors. Design equations. Non-ideal flow. Multiple reactors. Rate equations. Determination of kinetic data.

Planning				
Methodologies / tests	Competencies	Ordinary class hours	Student?s personal work hours	Total hours
Laboratory practice	A11 A19 A20 B2 B5 C2 C3	10	15	25
Guest lecture / keynote speech	A11 A15 A25 B2 C3	26	65	91
Problem solving	A11 A15 B2 C3	9	20.25	29.25
Mixed objective/subjective test	A11 A15 A25 B2	3	0	3
Personalized attention		1.75	0	1.75

(\*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
Methodologies	Description
Laboratory practice	Sesi3ns experimentais nas que os alumnos deber3n utilizar as montaxes prefijados para tentar comprobar o cumprimento de modelos te3ricos na pr3ctica.
Guest lecture / keynote speech	Sesi3ns nas que se explicar3n os contidos te3ricos de cada tema, estud3sen e resolveran varios exemplos e faranse alg3nos exercicios b3sicos de aplicaci3n, en grupos grandes.
Problem solving	Sesi3ns nas que os alumnos deber3n solucionar exercicios propostos dos diversos temas, en grupos pequenos.
Mixed objective/subjective test	Exame escrito que constase de preguntas de teor3a e/ou problemas.

Personalized attention
------------------------



Methodologies	Description
Laboratory practice	Os alumnos realizarán as prácticas de laboratorio coa axuda e atención personalizada do profesor de prácticas. As pautas a seguir explicásense antes de empezar as prácticas.

Assessment			
Methodologies	Competencies	Description	Qualification
Laboratory practice	A11 A19 A20 B2 B5 C2 C3	Puntuácese o traballo realizado no laboratorio e o informe final(competencias: A11, A15, A19, A20, A25, B2, B5, C3)	15
Guest lecture / keynote speech	A11 A15 A25 B2 C3	Participación en clase e resolución de exercicios (competencias: A11, A15, A25, B2, C2, C3)	5
Mixed objective/subjective test	A11 A15 A25 B2	Exame escrito (teoría e/ou problemas) (competencias: A11, A15, A25, B2)	80

Assessment comments
<ul style="list-style-type: none"> <li>- The work done in the laboratory will be taken into account as well as the report describing the results, corresponding to the analysis of data, and conclusions: 15% of the final score.</li> <li>- The assistance to the activities as well as exercises and problem solving: 5% of the final score.</li> <li>- Final examination: 80% of the final score.</li> <li>- The overall score will be the sum of the above described sub-scores. It will be considered that the student did not present the subject's exam if he/she did not go for the final examination.</li> <li>- In order to pass, the student has to obtain a minimum score of 5/10 in each of the three parts described above.</li> <li>- It will be necessary to attend the laboratory course in order to pass the subject.</li> <li>- The "matrícula de honor" will be rewarded to the students that achieved the maximum score in the first opportunity of evaluation. In the second opportunity, the same score will be maintained for the lab-course and the exercises (representing 15% and 5% of the final score) but it will be required to repeat the final written exam that will represent 80% of the final score. For successive academic courses, the teaching-learning process will have to start again, and the student will therefore have to repeat all activities and examinations for that new academic course.</li> </ul>

Sources of information	
<b>Basic</b>	<ul style="list-style-type: none"> <li>- COSTA LÓPEZ y col. (). Curso de Química Técnica: Introducción a los procesos, las operaciones unitarias y los fenómenos de transporte en la Ingeniería Química. Editorial Reverté, Barcelona</li> <li>- LEVENSPIEL, O., (). Ingeniería de las reacciones químicas. Ed. Reverté, Barcelona</li> <li>- COSTA NOVELLA y col. (). Ingeniería Química. Vol. 1. Conceptos generales. Edición Alhambra, Madrid</li> <li>- THOMPSON, E.V. &amp; CECKLER, W.H., (). Introducción a la Ingeniería Química. McGraw-Hill</li> <li>- FELDER, R.M. &amp; ROUSSEAU, R.W., (). Principios elementales de los procesos químicos. Addison- Wesley Iberoamericana, Wilmington</li> <li>- HIMMELBLAU, D.M., (). Principios y cálculos básicos de Ingeniería Química. C.E.C.S.A. México</li> </ul>
<b>Complementary</b>	

Recommendations
Subjects that it is recommended to have taken before
Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus
Other comments
&nbsp;



(\*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.