		Teaching	Guide		
	Identifyir	ng Data			2016/17
Subject (*)	Química Orgánica 1			Code	610G01026
Study programme	Grao en Química				'
	<u>'</u>	Descrip	otors		
Cycle	Period	Yea	r	Туре	Credits
Graduate	1st four-month period	Seco	nd	Obligatoria	6
Language	SpanishEnglish				
Teaching method	Face-to-face				
Prerequisites					
Department	Química Fundamental				
Coordinador	Peinador Veira, Carlos E-mail carlos.peinador@udc.es			@udc.es	
Lecturers	García Romero, Marcos Daniel	García Romero, Marcos Daniel		marcos.garcia1@udc.es	
	Ojea Cao, Vicente			vicente.ojea@u	dc.es
	Peinador Veira, Carlos			carlos.peinador	@udc.es
	Rodriguez Gonzalez, Jaime			jaime.rodriguez	@udc.es
	Ruiz Pita-Romero, Maria			maria.ruiz.pita-r	romero@udc.es
Web	campusvirtual.udc.es/moodle/				
General description	The course provides basics of Or	ganic Chemistry	for students of De	egree in Chemistry	

	Study programme competences / results
Code	Study programme competences / results
A1	Ability to use chemistry terminology, nomenclature, conventions and units
A4	Knowledge of main types of chemical reaction and characteristics of each
A6	Knowledge of chemical elements and their compounds, synthesis, structure, properties and reactivity
A9	Knowledge of structural characteristics of chemical and stereochemical compounds, and basic methods of structural analysis and
	research
A10	Knowledge of chemical kinetics, catalysis and reaction mechanisms
A14	Ability to demonstrate knowledge and understanding of concepts, principles and theories in chemistry
A15	Ability to recognise and analyse new problems and develop solution strategies
A21	Understanding of qualitative and quantitative aspects of chemical problems
B2	Effective problem solving
В3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
С3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life

Learning outcomes			
Learning outcomes	Study	/ progra	amme
	con	npetenc	es/
		results	
Understand basic concepts, principles and theories related to the Organic Chemistry	A1	B2	C1
	A4	В3	
	A6	B4	
	A9		
	A10		
	A14		
	A15		
	A21		

Use and get use to the Organic Chemistry Nomenclature, using main conventions and units	A1	B2	C1
	A6	В3	C3
	A9		
	A14		
Understand the caracteristics and main propierties of organic compounds.	A1	B2	
	A9	В3	
	A14		
	A21		
Adquire the expertise to solve estructural and synthetic problems in Organic Chemistry being able to relate functional groups	A1	B2	C3
and their reactivities.	A4	В3	
	A9	B4	
	A14		
	A15		
	A21		
Understand the most impoortant type of organic reactions, the mechanistic pathways and its stereochemical features.	A1	B2	C1
	A4	В3	СЗ
	A6	B4	
	A9		
	A10		
	A21		
Design and planning in Organic Synthesis. Relationships between spectroscopic data and functional groups in Organic	A1	B2	СЗ
Chemistry.	A4	В3	
	A9		
	A14		
	A15		
Adquire expertise in literature searches from Organic Chemistry sources.	A6	В3	C1
	A9	B4	C3
	A14		
The Organic Chemistry as a part of our lives	A14	B4	C1
	A15		C3
	A21		

Contents		
Topic	Sub-topic	
Chemical structure and chemical bonding in organic	Lewis Structures. Resonance Structures. Atomic and hybrid orbitals. Chemical	
molecules	structure and bonding on methane, ethene and ethyne.	
2. Organic reactions	Characteristics of the organic compounds. Types of reaction Mechanisms. Types of	
	reagents: acids and bases, oxidants and reductors, electrophilicity and nucleophilicity.	
	Classification of the Organic reactions. Thermodynamic and kinetic features of	
	Organic Reactions. Kinetic and thermodynamic control. Reaction intermediates.	
3. Alkanes and cycloalkanes. Reactions	Physical properties. n-Alkanes: Rotational barriers. Conformational isomerism.	
	Cycloakanes: ring strain.	
4. Stereochemistry	Optical activity. Chirality and enantiomers. Nomenclature. Diasteroisomers.	
	Stereoisomerism in cyclic molecules.	
5. NMR Spectroscopy	Basic principles of the NMR. The most important nuclei studied in Organic Chemistry.	
	The chemical shift, Spin-spin coupling: N+1 rule. Identification of Organic functional	
	groups by NMR.	
6. Reactions of Alkanes.	Halogenation. Pyrolisis. Combustion. Cracking.	

7. Alkyl Halides	Chemical structure and properties. Nucleophilic substition reactions (SN). Factors determining the SN mechanisms: substrate (structure of the alkyl group and nature of the leaving group, nucleophilicity of the reagent and solvent effects. Competitive proccesses in the SN reaction: transpositions and eliminations. Organometallic
	compounds. Reduction of alkyl halides.
8. Alcohols	Chemical structure. Acid-base propierties. Reactions through the O-H bond. Reactions through the C-O bond. Oxidation. Thiols.
9. Ethers	Chemical structure. Ether cleavage. Epoxides. Thioethers.
10. Amines	Structure and properties. Acid-base reactions. Alkylation of amines. Hoffman's elimination. Oxidation: Cope's elimination.

	Planning	g		
Methodologies / tests	Competencies /	Teaching hours	Student?s personal	Total hours
	Results	(in-person & virtual)	work hours	
Mixed objective/subjective test	A1 A4 A6 A9 A10 A14	5	0	5
	A15 A21 B3 B2 C1			
Seminar	A1 A4 A6 A9 A10	10	25	35
Guest lecture / keynote speech	A1 A4 A6 A9 A10	20	30	50
ICT practicals	A6 A9 A21 B2 B4 C3	10	10	20
Workshop	A1 A4 A6 A9 A10 B2	10	30	40
	B4			
Personalized attention		0	0	0

	Personalized attention		
Methodologies	Description		
Seminar			
ICT practicals			
Workshop			

		Assessment	
Methodologies Competencies /		Description	
	Results		
Mixed	A1 A4 A6 A9 A10 A14	Final written test.	70
objective/subjective	A15 A21 B3 B2 C1		
test			
Seminar	A1 A4 A6 A9 A10	The evaluation will consider the attendance, participation, and completion of a	5
		colection of practice tests of proposed problems.	
ICT practicals	A6 A9 A21 B2 B4 C3	The follow up and participation on the practical sessions will contribute a 5% to the	10
		final grading.	
		The students will have to deliver a written final report of the practicals. This report will	
		contribute another 5% to the evaluation.	
Workshop	A1 A4 A6 A9 A10 B2	Attendance, participation, and the quality of the written solutions in the take home	15
	B4	assigned problems will be evaluated. The students will deliver the written solutions	
		before of each presentation. Additionally the clarity and precision of explanations, as	
		well as the use of appropriate nomenclature for the compounds and reactions it will be	
		evaluated.	

Assessment comments

The assistance to the ICT practical is a necessary requirement for passing the course.

The final grade will be based on your performances in the all activities. The exam will make up 70%, seminar 5%, ICT practicals 10%, and workshop 15%. Each activity (except for seminar) should reach a minimum score of 4 out of 10. A final score of 5 out of 10 is needed in order to pass the entire course.

The scores obtained on the practicals, seminars and workshop classes will be preserved to the "second opportunity" of an academic course. Those students participating in less than a 70% of the planned activities will obtain an assessment of "not attended".

Students evaluated in the "second chance" shall be eligible for "class honors" if the maximum number of those marks for the corresponding course has not been fully covered in the "first opportunity".

Part-time dedication students with academic waiver exempting assistance, will be provided with a flexible schedule according with coordination time-table, available material and human resources. The remaining activities (seminars and workshop) will be evaluated in the final written test. Part-time dedication students unable to attend the ICT practicals, will be evaluated with complementary questions about the laboratory practices in the final writing test. This part will account for the 10% of the final grade.

	Sources of information
Basic	- K.P.C. VOLLHARDT and N.E. SCHORE (2007). Organic Chemistry (5 ^a ed). Barcelona, Omega
	- L.G. WADE, Jr. (2004). QUÍMICA ORGÁNICA (5ª ed). Madrid, Pearson Educación
	- E. QUIÑOÁ y R. RIGUERA (2004). CUESTIONES Y EJERCICIOS DE QUÍMICA ORGÁNICA (2ª ed). Madrid,
	McGraw-Hill
Complementary	

	Recommendations
	Subjects that it is recommended to have taken before
Química 1/610G01007	
Química 2/610G01008	
Química 3/610G01009	
Química 4/610G01010	
	Subjects that are recommended to be taken simultaneously



Subjects that continue the syllabus

Química Orgánica 2/610G01027

Ampliación de Química Orgánica/610G01028

Experimentación en Química Orgánica/610G01029

Química Orgánica Avanzada/610G01030

Other comments

We highly recommend as main sources of information the books: Wade 2004 and Vollhardt, 2007. We also recommend the reading of the main subjects of each topic, as preparation to the teacher's keynote lectures.

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.