



| Guía Docente | | | | |
|-----------------------|--|--------------------|--|----------|
| Datos Identificativos | | | | 2016/17 |
| Asignatura (*) | Elucidación de Mecanismos de Reacción | Código | 610500013 | |
| Titulación | | | | |
| Descriptorios | | | | |
| Ciclo | Período | Curso | Tipo | Créditos |
| Mestrado Oficial | 2º cuatrimestre | Primeiro | Optativa | 3 |
| Idioma | CastelánGalegoInglés | | | |
| Modalidade docente | Presencial | | | |
| Prerrequisitos | | | | |
| Departamento | Química Física e Enxeñaría Química 1 | | | |
| Coordinación | Canle López, Moisés | Correo electrónico | moises.canle@udc.es | |
| Profesorado | Canle López, Moisés Fernandez Perez, Maria Isabel | Correo electrónico | moises.canle@udc.es isabel.fernandez.perez@udc.es | |
| Web | | | | |
| Descrición xeral | <p>The contents of the subject "Elucidation of Reaction Mechanisms" are oriented to complement previous knowledge from the graduation studies. Usually, reaction mechanisms are proposed for chemical processes without any indication of the experimental evidences that lead to such mechanism instead of any other. This subject will show which are such evidences and how they can be obtained.</p> <p>It is not frequent to face the planification of a research into how chemical reactions take places. This subject will face this kind of problem from a practical point of view. There are a number of techniques, direct and indirect evidences that allow the elucidation of the mechanism of a chemical process.</p> <p>Chemical reactivity is central to changes in nature, and the recognition of the different reaction mechanisms is fundamental to control chemical process, from the kinetic, thermodynamic points of view or even from the point of view of the generated products.</p> | | | |

| Competencias / Resultados do título | |
|-------------------------------------|-------------------------------------|
| Código | Competencias / Resultados do título |

| Resultados da aprendizaxe | | | |
|--|--|---|---------------------------|
| Resultados de aprendizaxe | Competencias / Resultados do título | | |
| | To go deeper into the physical basis of chemical reactivity. | AM4 AM6 AM7 AM8 AM9 AM20 | BM1 BM2 |
| To expand the knowledge and ability to use experimental techniques to determine and measure chemical reactivity and its changes. | AM11 AM22 | BM7 | |
| To understand the different concepts and theories necessary to characterize chemical processes and their course. | AM4 AM7 AM9 | BM2 BM3 BM6 | CM1 CM3 CM9 CM11 |
| To be able to use different instruments that are frequently used for the charecterization of reaction mechanisms. | AM9 AM11 AM22 | BM3 BM7 | CM3 |



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|--|-----|------|------|
| To be able to use / apply acquired abilities and concepts for the resolution of practical examples.. | AM1 | BM2 | CM2 |
| | AM3 | BM3 | CM3 |
| | AM4 | BM4 | CM4 |
| | AM6 | BM5 | CM5 |
| | | BM6 | CM9 |
| | | | CM10 |
| | | CM11 | |

| Contidos | |
|--|---|
| Temas | Subtemas |
| Reaction media | Variables that influence chemical processes. Role of reaction medium in chemical processes |
| Reaction mechanisms | Classification of reaction mechanisms Kinetic and thermodynamic characteristics of the main reaction mechanisms |
| Experimental techniques for the elucidation of reaction mechanisms | Batch methods Continuous methods Techniques for the study of rapid and ultrarapid reactions |
| Chemical reactivity | Catalysis Kinetic isotope effects Linear free energy relationships (LFER) and quantitative structure-activity relationships |
| Photochemistry | General concepts Photochemical processes Photochemistry and photoreactivity |

| Planificación | | | | |
|--------------------------|---|---|-------------------------|--------------|
| Metodoloxías / probas | Competencias / Resultados | Horas lectivas (presenciais e virtuais) | Horas traballo autónomo | Horas totais |
| Sesión maxistral | A1 A4 A6 A7 A8 A9 A11 A20 A22 B1 B5 C10 | 10 | 20 | 30 |
| Estudo de casos | A3 B2 B3 B4 B6 B7 C2 C3 C4 C5 C9 C11 | 4 | 6 | 10 |
| Prácticas de laboratorio | B2 B4 B6 C1 C4 C5 | 15 | 18.75 | 33.75 |
| Proba obxectiva | A4 A6 A7 A8 A9 A11 A20 A22 B2 B3 B4 B6 | 1 | 0 | 1 |
| Atención personalizada | | 1 | 0 | 1 |

*Os datos que aparecen na táboa de planificación son de carácter orientativo, considerando a heteroxeneidade do alumnado

| Metodoloxías | |
|--------------------------|---|
| Metodoloxías | Descrición |
| Sesión maxistral | ? Two-hour sessions to present the masterlines of the subject, indicating the students the most relevant points to take into account when studying and recommending appropriate materials for a better comprehension. ? The students will have the audiovisual material available through the Moodle virtual platform. |
| Estudo de casos | ? Different real cases will be critically analyzed and discussed, in order to apply the acquired knowledge |
| Prácticas de laboratorio | ? Will take place in the laboratory, in the days and hours that will be announced. ? At the end of the practical lessons, the student will hand a report on the experimental project developed, and realize a short oral presentation analyzing the experimental part and the meaning of the obtained results. |



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| Proba obxectiva | ? There will be a short exam, that may include both theory and practice |
|-----------------|---|

Atención personalizada

| Metodoloxías | Descrición |
|---|--|
| Estudo de casos Prácticas de laboratorio | Will be carried out at the lecturers' offices, or at the Laboratory of Physical Chemistry I, according to the established timetable (consult for each lecturer). Proposed exercises, laboratory reports, etc. may be hand directly in these hours, solving any doubt or question about them. Doubt or questions with a simple and brief answer may be asked and answered through the Moodle virtual platform. More complicated topics will need an appointment. |

Avaliación

| Metodoloxías | Competencias / Resultados | Descrición | Cualificación |
|--------------------------|---|--|---------------|
| Estudo de casos | A3 B2 B3 B4 B6 B7 C2 C3 C4 C5 C9 C11 | Evaluation will be centered in the critical analysis of the proposed cases, as well as on the suggestion of alternative solutions. | 20 |
| Proba obxectiva | A4 A6 A7 A8 A9 A11 A20 A22 B2 B3 B4 B6 | May include short test or multiple choice questions or short problems / cases to analyze. | 40 |
| Prácticas de laboratorio | B2 B4 B6 C1 C4 C5 | Both the experimental design and the critical analysis of the obtained results will be evaluated. | 40 |
| Outros | | | |

Observacións avaliación

Para superar la asignatura habrá que asistir tanto a las prácticas de laboratorio como a las simulaciones.

Fontes de información

| | |
|------------------------------------|---|
| Bibliografía básica | <ul style="list-style-type: none">- H. Maskill (1985). The Physical Basis of Organic Reactivity. Oxford University Press- Study materials or reference to them will be accesible through the Moodle virtual platform. |
| Bibliografía complementaria | <ul style="list-style-type: none">- H. Maskill (Ed.), (2006). Investigating Organic Reaction Mechanisms . Blackwell Science- N. J. Turro; V. Ramamurthy; J.C. Scaiano (2009). Principles of Molecular Photochemistry. An Introduction. University Science Books- E.V. Anslyn, D.A. Dougherty (2006). Modern Physical Organic Chemistry. University Science Books |

Recomendacións

Materias que se recomenda ter cursado previamente

Materias que se recomenda cursar simultaneamente

Materias que continúan o temario

Observacións

A higher profit from this subject would require actualized knowledge of Physical Chemistry. It is strongly recommended to review the theoretical concepts introduced in the lessons through the resolution of questions and exercises, that will be proposed.

(*)A Guía docente é o documento onde se visualiza a proposta académica da UDC. Este documento é público e non se pode modificar, salvo casos excepcionais baixo a revisión do órgano competente dacordo coa normativa vixente que establece o proceso de elaboración de guías