

	Identifying Da	ita		2017/18	
Subject (*)	Experimental Organic Chemistry		Code	610G01029	
Study programme	Grao en Química		· · · · · · · · · · · · · · · · · · ·		
		Descriptors			
Cycle	Period	Year	Туре	Credits	
Graduate	2nd four-month period	Third	Obligatoria	6	
Language	SpanishEnglish			,	
Teaching method	Face-to-face				
Prerequisites					
Department	Química				
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	Study programme competences / results
Code	Study programme competences / results
A1	Ability to use chemistry terminology, nomenclature, conventions and units
A9	Knowledge of structural characteristics of chemical and stereochemical compounds, and basic methods of structural analysis and
	research
A10	Knowledge of chemical kinetics, catalysis and reaction mechanisms
A15	Ability to recognise and analyse new problems and develop solution strategies
A17	Ability to work safely in a chemistry laboratory (handling of materials, disposal of waste)
A19	Ability to follow standard procedures and handle scientific equipment
A20	Ability to interpret data resulting from laboratory observation and measurement
A21	Understanding of qualitative and quantitative aspects of chemical problems
A22	Ability to plan, design and develop projects and experiments
A23	Critical standards of excellence in experimental technique and analysis
A26	Ability to follow standard laboratory procedures in relation to analysis and synthesis of organic and inorganic systems
B2	Effective problem solving
B3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life

Learning outcomes			
Learning outcomes			amme
		competences /	
		results	
Knowledge the characteristics and properties of organic compounds, their reactivity and the main reaction mechanisms,		B3	
including stereochemical aspects		B4	
	A23		



Design, plan and execute synthesis of organic molecules. Conducting processes of isolation, purification and characterization.	A15	B2	
Ability to manage the literature and finding specific information in organic chemistry.	A17		
	A21		
	A22		
	A26		
Knowledge of fundamental characteristics of organic compounds and the most important methods of preparation and	A9	B3	
structural determination of these compounds.	A17		
	A19		
	A20		
Carry out organic chemistry experiments independently and handling reagents safely. Manage scientific instrumentation in	A1	B2	C1
organic chemistry laboratory and interpret the results.	A9	B4	
	A10		
	A15		
	A17		
	A19		
	A20		
	A22		
Ability to manage literature, as well as a search of specific information in Organic Chemistry.	A15	B3	C3
	A22		
	1		

	Contents
Торіс	Sub-topic
Presentation	Methods, programmed activities and evaluation criteria
Carbonyl group.	Experiment 1a: Vainilline reduction with sodium borohydride.
Reduction reactions, synthesis of commercially interesting	Experiment 1b: Methyl diantilis synthesis.
products	
Alkenes, alkyl halides, alcohols and epoxides.	Experiment 2: Stereospecific synthesis of anti-2-bromo-1,2-diphenylethanol from
Electrophilic addition to unsaturated systems, bimilecular	trans-stilbene.
nucleophilic substitution ans rearrangenments.	
Aromatic compounds and electrophilic aromatic substitution.	Experiment 3: Synthesis of p-nitroaniline from aniline.
Introduction to protecting groups.	
Carboxylic acid derivatives.	Experiment 4a: Synthesis of ethyl acetate.
Nucleophilic substitution through addition-elimination.	Experiment 4b: Synthesis of isoamyl acetate.
Sustainable chemistry. Reactions without solvents.	Experiment 5: Synthesis of N-(2-hydroxy-3-methoxybenzyl)-N-p-tolylacetamide.
Carbonyl compounds and reactions in alpha position.	Experiment 6a: Synthesis of dibenzalacetone
	((E,E)-1,5-diphenyl-1,4-pentadien-3-one) from acetone and benzaldehyde through
	aldol condensation.
	Experiment 6b: Synthesis of ketone alfa, beta-unsaturated
	(6-etoxicarbonyl-3,5-diphenyl-2-cyclohexanone) through Michael reactions and aldol
	condensation.
Dienes. Diels-Alder reaction	Experiment 7: Synthesis of exo- and
	endo-7-oxabicyclo[2.2.1]hept-5-en-2,3-dicarboxy-N-phenylimide from
	N-phenylmaleimide
Polifunctional compounds.	Experiment 8a: Synthesis of benzylic acid from benzaldehyde.
Multistep synthesis	Experiment 8b: Synthesis of 3-methylcyclohexen-2-one trough Robinson annulation
	and decarboxylation
	Experiment 8c: Stereoselective reduction of benzoin adn synthesis of
	4,5-diphenyl-2,2-dimethyl-1,3-dioxolan
	Experiment 8d: Regioselective epoxydation of (R)-carvone.
	Experiment 8e:Synthesis of local anesthetic benzocaine (ethyl p-aminobenzoate).



Organophosphorous compounds.	Experiment 9: Synthesis of cynnamic acid through Wittig reaction.
Olefination reactions.	
Heterocyclic compounds.	Experiment 10a: Synthesis of 6-methylquinolin through Skraup reaction.
Synthesis. Green chemistry and pharmacologicaly interesting	Experiment 10b: Synthesis of 1,6-dihydropyridines through Hantzsch reaction in
heterocycles.	solventless conditions.
	Práctica 10c: Synthesis of Fischer indole: preparation of 1,2,3,4-tetrahydrocarbazole.
Carbohydrates.	Experiment 11a: Synthesis of beta- and alpha-D-glucose pentaacetates.
Kinetic and thermodinamic control. Protecting groups. Sugars	Experiment 11b: Synthesis of 2,3-O-isopropyliden-L-erithrose from L-arabinose
as chiral precursors.	
Amino acids and peptides	Experiment 12: Synthesis of methyl N-acetyl-L-prolyl-L-phenylalaninate from its amino
	acids.

Competencies / Results A1 A10 A15 A21 A22 A23 A26 B2 B3 C1 A1 A9 A10 A15 A20	Teaching hours (in-person & virtual) 2	Student?s personal work hours 0	Total hours
A1 A10 A15 A21 A22 A23 A26 B2 B3 C1			2
A23 A26 B2 B3 C1	2	0	2
AT A9 ATU AT5 A20	12	36	48
A23 A26 B2 B3 B4 C1			
C3			
A1 A4 A9 A15 A16	44	44	88
A17 A18 A19 A20			
A21 A22 A23 A24			
A26 B2 B3 B4 C1			
A1 A4 A9 A10 A15	2	8	10
A18 A19 A20 A21			
A22 A23 B2 B3 B4 C1			
	2	0	2
	C3 A1 A4 A9 A15 A16 A17 A18 A19 A20 A21 A22 A23 A24 A26 B2 B3 B4 C1 A1 A4 A9 A10 A15 A18 A19 A20 A21 A22 A23 B2 B3 B4 C1	C3 A1 A4 A9 A15 A16 44 A17 A18 A19 A20 A21 A22 A23 A24 A26 B2 B3 B4 C1 A1 A4 A9 A10 A15 A1 A4 A9 A10 A15 A18 A19 A20 A21 A22 A23 B2 B3 B4 C1 2	C3 A1 A4 A9 A15 A16 44 44 A17 A18 A19 A20 A21 A22 A23 A24 A26 B2 B3 B4 C1 44 A1 A4 A9 A10 A15 2 8 44 A18 A19 A20 A21 A22 A23 B2 B3 B4 C1 44 44

(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the stude

	Methodologies		
Methodologies	Description		
Introductory activities	A session is programmed in a only group in which students will be exposed to the teaching methodology, planned activities		
	and the evaluation criteria to be applied during the course program. Available resources will be presented on the website of		
	matter and the dates on which the experiences and interviews will be conducted for students to organize their previous work		
	indicated. Finally accurate information will be provided for students to start preparing for the first practice.		
Supervised projects	Tutored work includes assistance to 6 classroom sessions up to 2 hours, in which he will supervise and evaluate the work of		
	the independent student for the preparation of the labs. It will be held one interview per lab. Before the start of the interviews		
	students must have completed the previous work every practice in the laboratory notebook, which can be replaced in some		
	cases by a report of the preparatory work done to be delivered to the teacher. During interviews, the teacher will resolve the		
	doubts that may arise and will evaluate the work done.		
	The preparation work practices should include calculations, experimental procedures and mounts necessary for the		
	experience as well as an explanation of the mechanisms involved in the processes and solutions to questions of scripts to		
	follow.		



Laboratory practice	There will be 13 sessions of up to four hours of work, where students will do some of the planned experiments are scheduled.
	Prior to entering the laboratory, from a screenplay experience and bibliographic information available on the website of the
	subject, the student must work independently in the preparation of each experience way.
	During laboratory sessions, simultaneously with the completion of the experiments so, students must develop a laboratory
	notebook, which collect the calculations, the experimental procedures and the necessary setups. The teacher will review the
	laboratory notebook for each student in each practice
	After each practice, which may require several laboratory sessions, students must complete notebook with the results and
	conclusions, where the answers to the questions the script will include the structural elucidation of the compounds obtained
	and the data on its performance and purity.
Mixed	There will be a final written exam, in order to objectively assess the degree of assimilation and the applicability of the contents
objective/subjective	of the subject by students is scheduled in May.
test	

Personalized attention		
Description		
6 Interviews (with a total duration of 2 hours) are scheduled in which the teacher will carry out a follow-up, orientation and		
evaluation of non-face-to-face work done by the student for the preparation of laboratory sessions. Students should go to the		
interviews with a report of the preparation work done.		
In addition, the student can receive personalized attention on any aspect of the subject during the teacher's tutoring schedule		
In exceptional, objectivables and properly justified circumstances, the professor may fully or partly exempt any member of the student body to attend the ongoing evaluation process. Students that is in this circumstance must pass a specific test that leaves no doubt about achieving the powers of matter on two occasions.		
-		

		Assessment	
Methodologies	Competencies /	Description	Qualification
	Results		
Supervised projects	A1 A9 A10 A15 A20	In supervised self-made work done by the student for the autonomous preparation of	40
	A23 A26 B2 B3 B4 C1	the laboratory practices will be assessed. During tutorials students will have to expose	
	C3	part of the report's conclusions, the quality of the exhibition as well as active	
		participation in solving the problems will be assessed. The rating of this part includes	
		the assessment of laboratory notebook.	
Laboratory practice	A1 A4 A9 A15 A16	A continuous evaluation of the work in the laboratory where the interest and dedication	30
	A17 A18 A19 A20	of the students is taken into account, proper planning and organization of work,	
	A21 A22 A23 A24	respect for the safety and skill achieved in laboratory operations will be conducted.	
	A26 B2 B3 B4 C1		
Mixed	A1 A4 A9 A10 A15	In a joint test, the student must explain in writing and carried out similar to the	30
objective/subjective	A18 A19 A20 A21	practices in the laboratory experience program. From the data provided in the	
test	A22 A23 B2 B3 B4 C1	statement (description and amounts of the starting materials and products structure	
		synthesize) shall: (1) make all necessary calculations, (2) propose appropriate	
		experimental procedures for the preparation and purification compounds, (3) describe	
		the required assemblies and (4) propose reaction mechanisms that explain the	
		processes involved.	

Assessment comments



Attendance

at the presentation session, the laboratory practicum, the interviews and the examination are mandatory. To pass the course is necessary to obtain greater or equal to 5 out of 10 and a minimum return of 40%. Students whose average yield exceeds 4.9 points and that do not meet the minimum performance in any of the activities will be assessed as "unfit" and receive the grade of 4.9. We will only qualify as "not submitted" to students who have participated in activities that add less than 25% of the final grade. The marks obtained in interviews and in the labs will remain in the 2nd opportunity at July 2018. The 2nd opportunity, students may be submitted to a new evaluation of the mixed test to establish the 30% of the grade. According to the academic regulations, students are evaluated on the second occasion only choose honors if the maximum number of these not completed in full at the earliest opportunity With regard to the successive academic years, the process of teaching and learning, including assessment, refers to an academic year and thus begins again with a new academic year, including all activities and evaluation procedures that scheduled for that course. In exceptional, objectivables and properly justified circumstances, the professor may fully or partly exempt any member of the student body to attend the ongoing evaluation process. Students that is in this circumstance must pass a specific test that leaves no doubt about achieving the powers of matter on two occasions. A asistencia á sesión de presentación, as prácticas de laboratorio, ás entrevistas e ao exame son obrigatorias. Para superar a materia será necesario obter unha cualificación media maior ou igual a 5 puntos sobre 10 e un rendemento mínimo do 30% en cada unha das actividades. Os alumnos cuxo rendemento medio supere 4,9 puntos e que non alcancen o rendemento mínimo nalgunha das actividades, serán avaliados como "non aptos" e recibirán a cualificación de 4,9. Só outorgarase a cualificación de "non presentado" aos alumnos que realizasen menos do 25% do total das actividades avaliables que se programan na guía docente. As cualificacións obtidas nas entrevistas e nas prácticas de laboratorio manteranse na segunda oportunidade de xullo de 2015. Na segunda oportunidade, os alumnos poderán presentarse a unha nova avaliación da proba mixta para establecer o 30% da cualificación, na data e o horario establecida pola Xunta de Facultade. Os alumnos que opten pola nova avaliación deberán porse en contacto co profesor de maneira previa á realización do exercicio para coñecer o contido da práctica que deberán expor. De acordo coa normativa académica, os alumnos que sexan avaliados na segunda oportunidade só poderán optar a Matrícula de Honra se o número máximo destas non se completou na súa totalidade na primeira oportunidade Polo que respecta aos sucesivos cursos académicos, o proceso de ensino-aprendizaxe, incluída a avaliación, refírese a un curso académico e por tanto volve comezar cun novo curso académico, incluíndo todas as actividades e procedementos de avaliación que se programen para devandito curso.

Sources of information



Basic	- Rodríguez Yunta, M. J.; Gómez Contreras, F. (2008). Curso Experimental en Química Orgánica . Madrid. Síntesis.
	- Harwood, L. M.; Moody, C. J.; Percy, J. M. (1998). Experimental Organic Chemistry. Standard and microscale.
	Oxford. Blackwell Science.
	- Mohrig, J. R.; Hammond, C. N.; Morrill, T. C.; Neckers, D. C. Organic Chemistry: A Balanced Approac (1998).
	Experimental Organic Chemistry: A Balanced Approach Organic Chemistry: A Balanced Approach Macroscale and
	Microscale . New York. Freeman
	- Mohrig, J. R.; Hammond, C. N.; Schatz, P. F.; Morrill, T. C. (2003). Modern projects and experiments in organic
	chemistry miniscale and standard taper microscale . New York. Freeman
	- Martínez Grau, Mª A.; Csaky, A. G. (1998). Técnicas Experimentales en Síntesis Orgánica . Madrid. Síntesis.
Complementary	

 Recommendations

 Subjects that it is recommended to have taken before

 Organic Chemistry 1/610G01026

 Organic Chemistry 2/610G01027

 Intermediate Organic Chemistry/610G01028

 Subjects that are recommended to be taken simultaneously

 Subjects that continue the syllabus

 Advanced Organic Chemistry/610G01030

 Other comments

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.