



Teaching Guide				
Identifying Data				2019/20
Subject (*)	TERMODINÁMICA		Code	730G04014
Study programme	Grao en enxeñaría en Tecnoloxías Industriais			
Descriptors				
Cycle	Period	Year	Type	Credits
Graduate	1st four-month period	Second	Obligatory	6
Language	Spanish			
Teaching method	Face-to-face			
Prerequisites				
Department	Ciencias da Navegación e Enxeñaría MariñaEnxeñaría Naval e Industrial			
Coordinador	Calvo Diaz, Jose Ramon	E-mail	jose.ramon.calvo@udc.es	
Lecturers	Calvo Diaz, Jose Ramon Lamas Galdo, Isabel	E-mail	jose.ramon.calvo@udc.es isabel.lamas.galdo@udc.es	
Web	www.udc.es			
General description	Heat, work, and energy.			

Study programme competences	
Code	Study programme competences
A7	CR1 Coñecementos de termodinámica aplicada e transmisión de calor. Principios básicos e a súa aplicación á resolución de problemas de enxeñaría.
B1	CB1 Que os estudantes demostren posuír e comprender coñecementos nunha área de estudo que parte da base da educación secundaria xeral e adoita encontrarse a un nivel que, aínda que se apoia en libros de texto avanzados, inclúe tamén algúns aspectos que implican coñecementos procedentes da vanguarda do seu campo de estudo
B3	CB3 Que os estudantes teñan a capacidade de reunir e interpretar datos relevantes (normalmente dentro da súa área de estudo) para emitiren xuízos que inclúan unha reflexión sobre temas relevantes de índole social, científica ou ética
B5	CB5 Que os estudantes desenvolvan aquellas habilidades de aprendizaxe necesarias para emprenderen estudos posteriores cun alto grao de autonomía
B7	B5 Ser capaz de realizar unha análise crítica, avaliación e síntese de ideas novas e complexas
B9	B8 Adquirir unha formación metodolóxica que garanta o desenvolvemento de proxectos de investigación (de carácter cuantitativo e/ou cualitativo) cunha finalidade estratéxica e que contribúan a situarnos na vanguarda do coñecemento
C4	C6 Valorar criticamente o coñecemento, a tecnoloxía e a información dispoñible para resolver os problemas cos que deben enfrentarse.
C6	C8 Valorar a importancia que ten a investigación, a innovación e o desenvolvemento tecnolóxico no avance socioeconómico e cultural da sociedade.

Learning outcomes			
Learning outcomes			Study programme competences
Learn thermodynamics and heat transfer. Fundamentals and application to engineering.			A7 B1 C4 B3 C6 B5 B7 B9

Contents	
Topic	Sub-topic



The following blocks or chapters develop the contents established in the verification memory, which are:	Fundamentals Energy and conservation of energy Entropy Exergy Problems applied to engineering
1. Introduction to thermodynamics	Thermodynamics and energy Systems and control volumes Properties States Processes Energy and enthalpy Specific heat and thermal capacity Phases Ideal gases Temperature and Zeroth Law of Thermodynamics Density Pressure
2. Work, energy and the 1st Law of Thermodynamics (conservation of energy)	Energy Energy transfer by heat Energy transfer by work The First Law of Thermodynamics for closed systems, energy balance
3. Propiedades de una sustancia pura	Introduction Phase-change processes of pure substances Property diagrams Property tables Properties of incompressible substances Properties of ideal gases Reference states
4. Conservation of energy and 1st law of Thermodynamics	Introduction Conservation of mass in control volumes Conservation of energy in control volumes Examples
5. 2nd Law of Thermodynamics and introduction to thermodynamic cycles	Introduction Thermal energy reservoirs Thermodynamic cyclic devices: heat engines, refrigerators and heat pumps Kelvin-Planck and Clausius statements for the Second Law of Thermodynamics Maximum thermal efficiency of thermodynamic cyclic devices
6. Entropy	AClausius inequality Entropy Entropy tables Entropy diagrams Tds relations Entropy change of thermal energy reservoirs Entropy change of incompressible substances Entropy change of ideal gases Entropy generation Isentropic processes Entropy balance for closed systems and control volumes Entropy of the universe Isentropic efficiency of pumps, compressors, turbines and nozzles



Planning				
Methodologies / tests	Competencies	Ordinary class hours	Student?s personal work hours	Total hours
ICT practicals	A7 B1 B3 B5 B7 B9 C4 C6	22	44	66
Guest lecture / keynote speech	A7 B1 B3 B5 B7 B9 C4 C6	21	42	63
Introductory activities	A7 B1 B3 B5 B7 B9 C4 C6	10	0	10
Mixed objective/subjective test	A7 B1 B3 B5 B7 B9 C6 C4	4	6	10
Personalized attention		1	0	1

(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
Methodologies	Description
ICT practicals	Classes using computers
Guest lecture / keynote speech	Classes
Introductory activities	Partial exam with initial aspects
Mixed objective/subjective test	Exam/s

Personalized attention	
Methodologies	Description
Introductory activities	Personal attention will be provided by tutorials and e-mail.
Mixed objective/subjective test	Academic dispense is allowed. Students who request it must contact teacher to realize additional homework.
ICT practicals	

Assessment			
Methodologies	Competencies	Description	Qualification
Introductory activities	A7 B1 B3 B5 B7 B9 C4 C6	Partial exam with initial aspects.	10
Mixed objective/subjective test	A7 B1 B3 B5 B7 B9 C6 C4	Exam/s.	70
ICT practicals	A7 B1 B3 B5 B7 B9 C4 C6	Students may deliver some exercises.	20
Others			

Assessment comments

Students

who request academic dispense must realize other activities proposed by the teacher. The qualification is the same as the practice.

The evaluation criteria of the 2nd opportunity are the same as those of the 1st opportunity except that, in case of partial exams, the mark obtained in these will not be taken into account in the 2nd opportunity.

In order to pass it is necessary to obtain at least 4 in the final exam and 5 in the global score.

Sources of information

Basic	<ul style="list-style-type: none"> - Y. A. Çengel; M. A. Boles. (). Thermodynamics. McGraw-Hill - M. Moran y H. N Shapiro (). Fundamentos de Termodinámica Técnica. John Willey & Sons - J. Mª Sáiz Jabardo (). Introducción a la Termodinámica. - Y. A. Cengel (). Ecuaciones Diferenciales para Ingeniería y Ciencias. McGraw-Hill
Complementary	

Recommendations
Subjects that it is recommended to have taken before

CALCULUS/730G01101

PHYSICS I/730G01102

DIFFERENTIAL EQUATIONS/730G01110

MECHANICS/730G01118

Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus

FLUID MECHANICS/730G01119

Industrial Heat Transfer/730G03020

Fluid and Thermal Machines/730G03023

Other comments



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To help achieve a sustained immediate environment and meet the objective of action number 5: "Healthy and sustainable environmental and social teaching and research" of the "Green Campus Ferrol Action Plan";</p><p>The delivery of the documentary works that are made in this matter:</p><p> Will be requested in virtual format and / or computer support</p><p> It will be done through Moodle, in digital format without the need to print them</p><p> If it is necessary to make them on paper:</p><p> Plastics will not be used</p><p> Double-sided prints will be made.</p><p> Recycled paper will be used.</p><p> Printing of drafts will be avoided.</p><p> A sustainable use of resources and the prevention of negative impacts on the natural environment must be made</p><p> The importance of ethical principles related to the values ??of sustainability in personal and professional behaviors must be taken into account</p><p> Gender perspective is incorporated into the teaching of this subject (non-sexist language will be used, bibliography of authors of both sexes will be used, intervention in class of students will be encouraged ...)</p><p> Work will be done to identify and modify prejudices and sexist attitudes, and the environment will be influenced to modify them and promote values ??of respect and equality.</p><p> Discrimination situations must be detected and actions and measures will be

proposed to correct them.

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(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.