

Teaching Guide				
Identifying Data				2020/21
Subject (*)	Applied Coordination Chemistry		Code	610509110
Study programme	Mestrado Universitario en Investigación Química e Química Industrial (Plan 2020)			
Descriptors				
Cycle	Period	Year	Type	Credits
Official Master's Degree	2nd four-month period	First	Optional	3
Language	Spanish			
Teaching method	Face-to-face			
Prerequisites				
Department	Química			
Coordinador	Fernandez Lopez, Alberto A.	E-mail	alberto.fernandez@udc.es	
Lecturers	Fernandez Lopez, Alberto A. Platas Iglesias, Carlos	E-mail	alberto.fernandez@udc.es carlos.platas.iglesias@udc.es	
Web				
General description	Este curso enmárcase na especialidade de Química Sintética dedicado ao estudo da síntese e propiedades de diversas especies químicas. Dentro de dita especialidade escolléronse algúns tipos de compostos que, xa sexa polas súas especiais características, xa sexa polas súas aplicacións futuras, merecen un estudo detallado. Tendo isto en conta, a Química da Coordinación merece un estudo aparte, tanto polas particulares características dos compostos de coordinación como polos seus métodos de sínteses. Desde outro punto de vista, os chamados complexos preséntanse nunha variedade estrutural enorme que vai desde especies de dimensión molecular pasando por agregados supramoleculares, polímeros mono-, bi-, e tridimensionais, ata chegar ao chamados Metal Organic Frameworks (MOFs) que forman redes tridimensionais ordenadas. Dada esta enorme variedade estrutural, non é de sorprender que o número de propiedades e aplicacións que presentan sexa tamén diverso. Todas estas razóns xustifican a súa inclusión tanto na especialidade de Química sintética como nun Máster dedicado ao estudo da Química.			
Contingency plan	1. Modifications to the contents There are no changes in the contents 2. Methodologies *Teaching methodologies that are maintained Lecture, seminar and mixed test. *Teaching methodologies that are modified Thera are no modifications to the methodologies, except that the learning platforms Moodle and Teams, or any other used by UDC, will be used in synchronous or asynchronous mode. 3. Mechanisms for personalized attention to students Email or Teams (or another similar platform), at the request of students. 4. Modifications in the evaluation There are no modifications in the evaluation methodology or percentages assigned to the different methodologies. *Evaluation observations In case the lectures or seminars can not be conducted in synchronous mode the assistance and active participation in these activities will not be evaluated. The final mixed test will be taken by using Moodle, Teams or any another learning platform used by UDC. 5. Modifications to the bibliography or webgraphy There are no modifications.			

Study programme competences	
Code	Study programme competences
A1	Define concepts, principles, theories and specialized facts of different areas of chemistry.
A2	Suggest alternatives for solving complex chemical problems related to the different areas of chemistry.
A3	Innovate in the methods of synthesis and chemical analysis related to the different areas of chemistry



A8	Analyze and use the data obtained independently in complex laboratory experiments and relating them with the chemical, physical or biological appropriate techniques, including the use of primary literature sources
B1	Possess knowledge and understanding to provide a basis or opportunity for originality in developing and / or applying ideas, often within a research context
B2	Students should apply their knowledge and ability to solve problems in new or unfamiliar environments within broader (or multidisciplinary) contexts related to their field of study.
B3	Students should be able to integrate knowledge and handle complexity, and formulate judgments based on information that was incomplete or limited, include reflecting on social and ethical responsibilities linked to the application of their knowledge and judgments.
B4	Students should be able to communicate their conclusions, and the knowledge and the reasons that support them to specialists and non-specialists in a clear and unambiguous manner
B7	Identify information from scientific literature by using appropriate channels and integrate such information to raise and contextualize a research topic
B10	Use of scientific terminology in English to explain the experimental results in the context of the chemical profession
B12	Being able to work in a team and adapt to multidisciplinary teams.
C1	CT1 - Elaborar, escribir e defender publicamente informes de carácter científico e técnico
C3	CT3 - Traballar con autonomía e eficiencia na práctica diaria da investigación ou da actividade profesional.
C4	CT4 - Apreciar o valor da calidade e mellora continua, actuando con rigor, responsabilidade e ética profesional.

Learning outcomes			
Learning outcomes		Study programme competences	
Students will be able to design new routes to prepare and isolate coordination compounds.	AC1	BC1	CC1
	AC2	BC2	CC3
	AC3	BC3	
	AC8	BC4	
		BC7	
Students will be able to identify the chirality in mononuclear coordination complexes and identify its origin		BC10	
	AC1	BC1	CC4
	AC2	BC2	
	AC3	BC3	
	AC8	BC7	
Students will be able to describe the factors that imply activation small molecules after coordination to metal centres and their applications.		BC10	
	AC1	BC1	
	AC2	BC2	
	AC3	BC3	
	AC8	BC7	
		BC10	
		BC12	

Contents	
Topic	Sub-topic
Structural properties of coordination compounds.	Structural properties in coordination compounds.
Characterization techniques	Characterization techniques.
	- Spectroscopic techniques
	- Spectrometric techniques
	- Diffraction techniques
	- Techniques based on magnetic properties
	- Other techniques



Activation of small molecules by coordination compounds	Structure and bonding in dioxygen and dinitrogen complexes: bonding modes. Synthesis of dioxygen and dinitrogen complexes. Structural characterization of dioxygen and dinitrogen complexes: NMR and IR spectroscopy, mass spectrometry, single crystal X-ray diffraction. Synthetic applications of dioxygen and dinitrogen complexes. Future perspectives.
Coordination compounds with applications in medicine: therapeutic and diagnose tools	Coordination compounds with applications in medicine: therapeutic and diagnose tools - Antitumor metalodrugs. - Types of complexes according to the metal centre and the structure - Therapeutic mechanism. - Synthetic methods. Application of metal complexes as radiopharmaceuticals. - Properties and characteristics of the ligands and the radioisotope - Diagnostic, (PET, SPEC) and therapeutic techniques. Contrast agents in magnetic resonance imaging (MRI) Proton relaxation agents T1 (Gd³⁺, Mn²⁺ y Fe³⁺) y T2. Parameters affecting the efficiency. - Agents based on saturation transfer by chemical exchange (CEST). - Agents based on other nuclei (, ³¹P). - Hyperpolarization.
Coordination compounds in the design of new materials: coordination polymers and MOFs. Properties and applications.	Types of systems arising from the metal-ligand association. Isolated and extensive associations (coordination polymers and MOFs) Main structural characteristics and characterization. Synthetic strategies and general characterization techniques. Properties and applications.

Planning				
Methodologies / tests	Competencies	Ordinary class hours	Student?s personal work hours	Total hours
Seminar	A1 A2 A3 B1 B2 B3 B4 B7 B10 B12 C1 C3 C4	7	21	28
Mixed objective/subjective test	A1 A2 A3 A8 B1 B10	2	18	20
Guest lecture / keynote speech	A2 A3 B3 B7 B12 C4	12	13	25
Personalized attention		2	0	2
(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.				

Methodologies	
Methodologies	Description
Seminar	Interactive sessions related to the different subjects with debates and exchange of opinions with students. Resolution of practical exercises (problems, test questions, interpretation and processing of information, evaluation of scientific publications, etc.)
Mixed objective/subjective test	Tests designed to gauge the achievement of both theoretical and practical knowledge and the acquisition of skills and attitudes.
Guest lecture / keynote speech	The lectures in which the contents of the subject will be explained with the assistance of illustrative examples. The class slides will be available, prior to the class. In some cases, if the number of students and their characteristics are adequate complementary methodologies as, for example, the case study or analyses of bibliographic sources might be used. The active participation of students will be encouraged. Attendance to lectures is not compulsory by highly advisable.



Personalized attention

Methodologies	Description
Seminar Mixed objective/subjective test	Individual or small group tutorials designed to support the students in the learning process. The student may attend to the tutorial freely to solve any doubts related to the subject. Tutorials are particularly important to solve the proposed problems or the assigned tasks. Students granted the ?part time? condition will be given special attention through the tutorials which will be dedicated specifically to the solution of the problem sheets.

Assessment

Methodologies	Competencies	Description	Qualification
Guest lecture / keynote speech	A2 A3 B3 B7 B12 C4	Continuous assessment of students through questions made during the classes. Attendance and active participation will also be assessed	10
Seminar	A1 A2 A3 B1 B2 B3 B4 B7 B10 B12 C1 C3 C4	Resolution of problems and study cases. Attendance will also be assessed as well as the active participation to de different activities.	35
Mixed objective/subjective test	A1 A2 A3 A8 B1 B10	Mixed test exam with questions and problems related to class contents.	55

Assessment comments

The ?no presentado? mark will be granted to those students who do not attend to the mixed text.

Students

granted the ?part time student? condition will be assessed exclusively by the mixed text marks (100% of the subject mark)

Sources of information

Basic	- J. Rivas Gispert (2000). Química de Coordinación. Omega - J. Rivas Gispert (2008). Coordination Chemistry . Weinheim: Wiley-VCH
Complementary	- Patrick L. Holland. Metal?dioxygen and metal?dinitrogen complexes: where are the electrons? Dalton Trans. , 2010, 39 , 5415?5425. - Michael P. Shaver, Michael D. Fryzuk. Activation of Molecular Nitrogen: Coordination, Cleavage andFunctionalization of N2 Mediated By Metal Complexes. Adv. Synth. Catal.2003, 345 , 1061- 1076 - Hiromasa Tanaka, Yoshiaki Nishibayashi, and Kazunari Yoshizawa, Interplay between Theory and Experiment forAmmonia Synthesis Catalyzed by Transition Metal Complexes, Acc. Chem. Res.2016, 49, 987?995.- Serenella Medici, Massimiliano Peana,Valeria Marina Nurchi, Joanna I. Lachowicz,Guido Crisponi, Maria AntoniettaZoroddu. Noble metalsin medicine: Latest advances. CoordinationChemistry Reviews, 2015, 284, 329?350.- A. Merbach, L. Helm and E. Tóth, The Chemistry of Contrast Agents inMedical Magnetic Resonance Imaging: Second Edition , John Wiley & Sons,Chichester, 2013. - Eric W. Price and Chris Orvig. Matchingchelators to radiometals for radiopharmaceuticals. Chem. Soc. Rev., 2014, 43,260-290.- - Stuart R. Batten, Neil R. Champness, Xiao-MingChen, Javier Garcia-Martinez, Susumu Kitagawa, Lars Öhrström, MichaelO?Keeffe7, Myunghyun Paik Suh, and Jan Reedijk. Terminology of metal?organic frameworks andcoordination polymers (IUPAC Recommendations 2013) . Pure Appl. Chem., 2013, 85, 1715?1724. - - Bradley J. Holliday and Chad A. Mirkin, Strategies for the Construction of Supramolecular Compounds throughCoordination Chemistry , Angew. Chem. Int. Ed. 2001, 40, 2022-2043. - Shin-ichiroNoro, Hitoshi Miyasaka, Susumu Kitagawa, Tatsuo Wada,Takashi Okubo, MasahiroYamashita, and Tadaoki Mitani. FrameworkControl by a Metalloligand Having Multicoordination Ability: New SyntheticApproach for Crystal Structures and Magnetic Properties . Inorg.Chem. 2005, 44, 133-146.



Recommendations

Subjects that it is recommended to have taken before

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Advanced Structural Determination/610509103

Subjects that are recommended to be taken simultaneously

Subjects that continue the syllabus

Other comments

The student must know the basic principles of coordination chemistry as, for example, the definition of coordination compound and its components, as well as the bonding theories used to describe this type of compounds

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.