



Teaching Guide

Teaching Guide				
Identifying Data				2020/21
Subject (*)	Thermodynamics		Code	730G05015
Study programme	Grao en Enxeñaría Naval e Oceánica			
Descriptors				
Cycle	Period	Year	Type	Credits
Graduate	1st four-month period	Second	Obligatory	6
Language	Spanish			
Teaching method	Hybrid			
Prerequisites				
Department	Ciencias da Navegación e Enxeñaría MariñaEnxeñaría Naval e Industrial			
Coordinador	Calvo Diaz, Jose Ramon	E-mail	jose.ramon.calvo@udc.es	
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Web	www.udc.es			
General description	Heat, work, and energy.			
Contingency plan	<div>1. Modifications to the contents</div> <div>No modifications.</div> <div>2. Methodologies</div> <div>*Teaching methodologies that are maintained</div> <div>No modifications.</div> <div>*Teaching methodologies that are modified</div> <div>No modifications.</div> <div>3. Mechanisms for personalized attention to students</div> <div>E-mail, moodle and teams. These will be consulted every dary.</div> <div>4. Modifications in the evaluation</div> <div>No modifications.</div> <div>*Evaluation observations:</div> <div>The exams will take place online.</div> <div>5. Modifications to the bibliography or webgraphy</div> <div>No modifications.</div>			

Study programme competences

Code	Study programme competences
A14	Knowledge of the applied thermodynamics and of the transmission of the heat.
B2	That the students know how to apply its knowledge to its work or vocation in a professional way and possess the competences that tend to prove itself by the elaboration and defense of arguments and the resolution of problems in its area of study
B3	That the students have the ability to bring together and to interpret relevant data (normally in its area of study) to emit judgments that include a reflection on relevant subjects of social, scientific or ethical kind
B4	That the students can transmit information, ideas, problems and solutions to a public as much specialized as not specialized
B5	That the students developed those skills of learning necessary to start subsequent studies with a high degree of autonomy
B6	Be able to carrying out a critical analysis, evaluation and synthesis of new and complex ideas.



C1	Using the basic tools of the technologies of the information and the communications (TIC) necessary for the exercise of its profession and for the learning throughout its life.
C4	Recognizing critically the knowledge, the technology and the available information to solve the problems that they must face.

Learning outcomes			
Learning outcomes		Study programme competences	
Model and calculate systems and processes related to the employment and generation of energy		A14	B2 B3 B4 B5 B6
			C1 C4

Contents	
Topic	Sub-topic
The following blocks or chapters develop the contents established in the verification memory, which are:	Introduction Conservation of energy Properties of pure substances 2nd law Practical applications
1. Introduction to thermodynamics	Thermodynamics and energy Systems and control volumes Properties States Processes Energy and enthalpy Specific heat and thermal capacity Phases Ideal gases Temperature and zeroth law of thermodynamics Density Pressure
2. Work, energy and the 1st law of thermodynamics (conservation of energy)	Energy Energy transfer by heat Energy transfer by work The first law of thermodynamics for closed systems, energy balance
3. Properties of pure substances	Introduction Phase-change processes of pure substances Property diagrams Property tables Properties of incompressible substances Properties of ideal gases Reference states
4. Conservation of energy and 1st law of thermodynamics	Introduction Conservation of mass in control volumes Conservation of energy in control volumes Examples



5. Thermodynamic cycles and introduction to the 2nd law of thermodynamics	<p>Introduction</p> <p>Thermal energy reservoirs</p> <p>Thermodynamic cyclic devices: heat engines, refrigerators and heat pumps</p> <p>Kelvin-Planck and Clausius statements for the second law of thermodynamics</p> <p>Maximum thermal efficiency of thermodynamic cyclic devices</p>
6. Entropy	<p>Clausius inequality</p> <p>Entropy</p> <p>Entropy tables</p> <p>Entropy diagrams</p> <p>T-ds relations</p> <p>Entropy change of thermal energy reservoirs</p> <p>Entropy change of incompressible substances</p> <p>Entropy change of ideal gases</p> <p>Entropy generation</p> <p>Isentropic processes</p> <p>Entropy balance for closed systems and control volumes</p> <p>Entropy of the universe</p> <p>Isentropic efficiency of pumps, compressors, turbines and nozzles</p>

Planning				
Methodologies / tests	Competencies	Ordinary class hours	Student's personal work hours	Total hours
ICT practicals	A14 B2 B3 B4 B5 B6 C1 C4	20	20	40
Guest lecture / keynote speech	A14 B2 B3 B4 B5 B6 C1 C4	30	30	60
Problem solving	A14 B2 B3 B4 B5 B6 C4 C1	20	20	40
Mixed objective/subjective test	A14 B2 B3 B4 B5 B6 C1 C4	0	9	9
Personalized attention		1	0	1
(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.				

Methodologies	
Methodologies	Description
ICT practicals	Classes using software
Guest lecture / keynote speech	Classes
Problem solving	Classes about problem solving
Mixed objective/subjective test	Exam/s

Personalized attention	
Methodologies	Description
Mixed objective/subjective test	Attention will be provided by personalized attention, e-mail and Teams.
Problem solving	Academic dispense is allowed. Students who request it must contact teacher to realize additional homework.
ICT practicals	



Assessment			
Methodologies	Competencies	Description	Qualification
Mixed objective/subjective test	A14 B2 B3 B4 B5 B6 C1 C4	Exam/s.	70
ICT practicals	A14 B2 B3 B4 B5 B6 C1 C4	Students may deliver some exercises.	30
Others			

Assessment comments
<p>Students who request academic dispense must realize other activities proposed by the teacher. The qualification is the same as the practice.</p> <p>The evaluation criteria of the 2nd opportunity are the same as those of the 1st opportunity except that, in case of partial exams, the mark obtained in these will not be taken into account in the 2nd opportunity.</p> <p>In order to pass it is necessary to obtain at least 4 in the final exam and 5 in the global score.</p>

Sources of information	
Basic	<ul style="list-style-type: none">- Y. A. Çengel; M. A. Boles. (). Thermodynamics. McGraw-Hill- M. Moran y H. N Shapiro (). Fundamentos de Termodinámica Técnica. John Wiley & Sons- J. Mª Sáiz Jabardo (). Introducción a la Termodinámica.- Y. A. Çengel (). Ecuaciones Diferenciales para Ingeniería y Ciencias. McGraw-Hill
Complementary	

Recommendations
Subjects that it is recommended to have taken before
CALCULUS/730G01101 PHYSICS I/730G01102 DIFFERENTIAL EQUATIONS/730G01110 MECHANICS/730G01118
Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus
FLUID MECHANICS/730G01119 Industrial Heat Transfer/730G03020 Fluid and Thermal Machines/730G03023
Other comments

proposed to correct them.

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