



Teaching Guide

Identifying Data					2020/21
Subject (*)	Thermodynamics	Code	730G05015		
Study programme	Grao en Enxeñaría Naval e Oceánica				
Descriptors					
Cycle	Period	Year	Type	Credits	
Graduate	1st four-month period	Second	Obligatory	6	
Language	Spanish				
Teaching method	Hybrid				
Prerequisites					
Department	Ciencias da Navegación e Enxeñaría MariñaEnxeñaría Naval e Industrial				
Coordinador	Calvo Díaz, Jose Ramon	E-mail	jose.ramon.calvo@udc.es		
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Web	www.udc.es				
General description	Heat, work, and energy.				
Contingency plan	<p>1. Modifications to the contents No modifications.</p> <p>2. Methodologies *Teaching methodologies that are maintained No modifications. *Teaching methodologies that are modified No modifications.</p> <p>3. Mechanisms for personalized attention to students E-mail, moodle and teams. These will be consulted every dary.</p> <p>4. Modifications in the evaluation No modifications. *Evaluation observations: The exams will take place online.</p> <p>5. Modifications to the bibliography or webgraphy No modifications.</p>				

Study programme competences

Code	Study programme competences
A14	Knowledge of the applied thermodynamics and of the transmission of the heat.
B2	That the students know how to apply its knowledge to its work or vocation in a professional way and possess the competences that tend to prove itself by the elaboration and defense of arguments and the resolution of problems in its area of study
B3	That the students have the ability to bring together and to interpret relevant data (normally in its area of study) to emit judgments that include a reflection on relevant subjects of social, scientific or ethical kind
B4	That the students can transmit information, ideas, problems and solutions to a public as much specialized as not specialized
B5	That the students developed those skills of learning necessary to start subsequent studies with a high degree of autonomy
B6	Be able to carrying out a critical analysis, evaluation and synthesis of new and complex ideas.



C1	Using the basic tools of the technologies of the information and the communications (TIC) necessary for the exercise of its profession and for the learning throughout its life.
C4	Recognizing critically the knowledge, the technology and the available information to solve the problems that they must face.

Learning outcomes			
Learning outcomes			Study programme competences
Model and calculate systems and processes related to the employment and generation of energy			A14 B2 B3 B4 B5 B6 C1 C4

Contents	
Topic	Sub-topic
The following blocks or chapters develop the contents established in the verification memory, which are:	Introduction Conservation of energy Properties of pure substances 2nd law Practical applications
1. Introduction to thermodynamics	Thermodynamics and energy Systems and control volumes Properties States Processes Energy and enthalpy Specific heat and thermal capacity Phases Ideal gases Temperature and zeroth law of thermodynamics Density Pressure
2. Work, energy and the 1st law of thermodynamics (conservation of energy)	Energy Energy transfer by heat Energy transfer by work The first law of thermodynamics for closed systems, energy balance
3. Properties of pure substances	Introduction Phase-change processes of pure substances Property diagrams Property tables Properties of incompressible substances Properties of ideal gases Reference states
4. Conservation of energy and 1st law of thermodynamics	Introduction Conservation of mass in control volumes Conservation of energy in control volumes Examples



5. Thermodynamic cycles and introduction to the 2nd law of thermodynamics	<p>Introduction</p> <p>Thermal energy reservoirs</p> <p>Thermodynamic cyclic devices: heat engines, refrigerators and heat pumps</p> <p>Kelvin-Planck and Clausius statements for the second law of thermodynamics</p> <p>Maximum thermal efficiency of thermodynamic cyclic devices</p>
6. Entropy	<p>Clausius inequality</p> <p>Entropy</p> <p>Entropy tables</p> <p>Entropy diagrams</p> <p>T-ds relations</p> <p>Entropy change of thermal energy reservoirs</p> <p>Entropy change of incompressible substances</p> <p>Entropy change of ideal gases</p> <p>Entropy generation</p> <p>Isentropic processes</p> <p>Entropy balance for closed systems and control volumes</p> <p>Entropy of the universe</p> <p>Isentropic efficiency of pumps, compressors, turbines and nozzles</p>

Planning				
Methodologies / tests	Competencies / Results	Teaching hours (in-person & virtual)	Student?s personal work hours	Total hours
ICT practicals	A14 B2 B3 B4 B5 B6 C1 C4	20	20	40
Guest lecture / keynote speech	A14 B2 B3 B4 B5 B6 C1 C4	30	30	60
Problem solving	A14 B2 B3 B4 B5 B6 C4 C1	20	20	40
Mixed objective/subjective test	A14 B2 B3 B4 B5 B6 C1 C4	0	9	9
Personalized attention		1	0	1

(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
Methodologies	Description
ICT practicals	Classes using software
Guest lecture / keynote speech	Classes
Problem solving	Classes about problem solving
Mixed objective/subjective test	Exam/s

Personalized attention	
Methodologies	Description
Mixed objective/subjective test Problem solving ICT practicals	<p>Attention will be provided by personalized attention, e-mail and Teams.</p> <p>Academic dispense is allowed. Students who request it must contact teacher to realize additional homework.</p>



Assessment

Methodologies	Competencies / Results	Description	Qualification
Mixed objective/subjective test	A14 B2 B3 B4 B5 B6 C1 C4	Exam/s.	70
ICT practicals	A14 B2 B3 B4 B5 B6 C1 C4	Students may deliver some exercises.	30
Others			

Assessment comments

Students

who request academic dispense must realize other activities proposed by the teacher. The qualification is the same as the practice.

The evaluation criteria of the 2nd opportunity are the same as those of the 1st opportunity except that, in case of partial exams, the mark obtained in these will not be taken into account in the 2nd opportunity.

In order to pass it is necessary to obtain at least 4 in the final exam and 5 in the global score.

Sources of information

Basic	<ul style="list-style-type: none"> - Y. A. Çengel; M. A. Boles. (). Thermodynamics. McGraw-Hill - M. Moran y H. N Shapiro (). Fundamentos de Termodinámica Técnica. John Willey & Sons - J. M^º Sáiz Jabardo (). Introducción a la Termodinámica. - Y. A. Cengel (). Ecuaciones Diferenciales para Ingeniería y Ciencias. McGraw-Hill
Complementary	

Recommendations

Subjects that it is recommended to have taken before

CALCULUS/730G01101
PHYSICS I/730G01102
DIFFERENTIAL EQUATIONS/730G01110
MECHANICS/730G01118

Subjects that are recommended to be taken simultaneously

Subjects that continue the syllabus

FLUID MECHANICS/730G01119
Industrial Heat Transfer/730G03020
Fluid and Thermal Machines/730G03023

Other comments



To help achieve a sustained immediate environment and meet the objective of action number 5:

"Healthy and sustainable environmental and social teaching and research" of the "Green Campus Ferrol Action Plan":

The delivery of the documentary works that are made in this matter:

Will be requested in virtual format and / or computer support

It will be done through Moodle, in digital format without the need to print them

If it is necessary to make them on paper:

Plastics will not be used

Double-sided prints will be made.

Recycled paper will be used.

Printing of drafts will be avoided.

A sustainable use of resources and the prevention of negative impacts on the natural environment must be made

The importance of ethical principles related to the values of sustainability in personal and professional behaviors must be taken into account

Gender perspective is incorporated into the teaching of this subject (non-sexist language will be used, bibliography of authors of both sexes will be used, intervention in class of students will be encouraged ...)

Work will be done to identify and modify prejudices and sexist attitudes, and the environment will be influenced to modify them and promote values of respect and equality.

Discrimination situations must be detected and actions and measures will be proposed to correct them.

(*The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.