

		Teachin	g Guide		
	Identifying Data 2021/22				
Subject (*)	Advanced Materials Characterization Techniques Code			610509121	
Study programme	Mestrado Universitario en Investiga	ación Química	a e Química Industri	al (Plan 2020)	
		Descr	riptors		
Cycle	Period	Ye	ear	Туре	Credits
Official Master's Degree	e 2nd four-month period	Fi	rst	Optional	3
Language	SpanishGalicianEnglish				
Teaching method	Face-to-face				
Prerequisites					
Department	Departamento profesorado máster	Química			
Coordinador	Sanchez Andujar, Manuel		E-mail	m.andujar@udc	.es
Lecturers	Sanchez Andujar, Manuel		E-mail	m.andujar@udc	.es
Web					
General description	This course includes a description	of the fundam	nentals and main ap	plications of several of	characterization techniques widely
	used in Materials Science and not	previously tre	ated in the compuls	ory subject "Materials	Characterization Techniques and
	Biointerphases" (module M1). The	se contents a	re important to com	plete the training in th	is module M5 -Nanoquímica and
	New Materials- and to have a more	e complete vis	sion of the technique	es of characterization	of materials and nanomaterials.
Contingency plan	In principle, the contents will be ke	pt in their enti	rety. However, if ne	cessary and for reasc	ons of force majeure, a more
	general presentation of the content	ts can be cho	sen, but in any case	all the most relevant	aspects of the subject will be
	covered.				

	Study programme competences / results
Code	Study programme competences / results
A1	Define concepts, principles, theories and specialized facts of different areas of chemistry.
A2	Suggest alternatives for solving complex chemical problems related to the different areas of chemistry.
A9	Promote innovation and entrepreneurship in the chemical industry and in research.
B2	Students should apply their knowledge and ability to solve problems in new or unfamiliar environments within broader (or multidisciplinary)
	contexts related to their field of study.
B3	Students should be able to integrate knowledge and handle complexity, and formulate judgments based on information that was
	incomplete or limited, include reflecting on social and ethical responsibilities linked to the application of their knowledge and judgments.
B5	Students must possess learning skills to allow them to continue studying in a way that will have to be largely self-directed or autonomous.
B8	Evaluate responsibility in the management of information and knowledge in the field of Industrial Chemistry and Chemical Research
B10	Use of scientific terminology in English to explain the experimental results in the context of the chemical profession
B11	Apply correctly the new technologies to gather and organize the information to solve problems in the professional activity.
C1	CT1 - Elaborar, escribir e defender publicamente informes de carácter científico e técnico
C3	CT3 - Traballar con autonomía e eficiencia na práctica diaria da investigación ou da actividade profesional.
C4	CT4 - Apreciar o valor da calidade e mellora continua, actuando con rigor, responsabilidade e ética profesional.

Learning outcomes			
Learning outcomes	Study programme		amme
	competences /		es/
		results	
- The student will obtain an overview of the advanced techniques of morphological, structural and microstructural	AC1	BC2	CC1
characterization.	AC2	BC3	CC3
- The student will learn the advantages and limitations of each one of the characterization technique.	AC9	BC5	CC4
- When you need to characterize a material, the student will be able to discern what are the characterization techniques that		BC8	
better fit your needs / possibilities.		BC10	
		BC11	



	Contents	
Topic Sub-topic		
Theme 1.	Introduction to microscopic techniques.	
microscopic techniques	Optical microscopies (fluorescence and confocal), electronic microscopies (TEM,	
	SEM, STEM, electron diffraction), scanning probe microscopies (STM, AFM).	
Theme 2.	Introduction to diffractometric techniques.	
diffractometric techniques	X-ray and synchrotron diffraction, neutron diffraction	
Theme 3.	electronic spectroscopic techniques. (EDXS, EELS)	
spectroscopic techniques.	electron paramagnetic resonance (EPR)	
Theme 4:	Physical adsorption of gases, specific surface area, pore size distribution.	
Characterization of porous materials		
Tema 5:	Single particle (SP-ICP-MS), and hybrid techniques (HPLC-ICP-MS, FFF-ICP-MS	
Atomic mass spectrometry techniques		

	Planning	g		
Methodologies / tests	Competencies /	Competencies / Teaching hours		Total hours
	Results	(in-person & virtual)	work hours	
Guest lecture / keynote speech	A1 A2 A9	12	0	12
Seminar	A1 B2 B3 B5 B8	7	0	7
Problem solving	A1 A2 A9 B2 B10 B11	0	24	24
	C1 C4			
Document analysis	C3 C4	0	12	12
Objective test	A1 A2 A9 B2 B3 B5	1	18	19
	B8 B10 B11 C1			
Personalized attention		1	0	1

(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

	Methodologies		
Methodologies	Description		
Guest lecture /	Theoretical classes. Magisterial lessons (with the use of blackboard and computer), complemented with the tools of virtual		
keynote speech	teaching		
Seminar	Practical seminars conducted by teachers of the Master, or invited professionals from companies, the Administration or other		
	universities. Interactive sessions related to the subjects with discussions and exchange of points of view with the students		
Problem solving	Resolution of practical exercises (problems, quizzes, processing and interpretation of information, evaluation of scientific		
	publications, etc.).		
Document analysis	Personal study based on different sources of information.		
Objective test	Preparation of the different tests for verification of obtaining both theoretical and practical knowledges, and the acquisition of		
	skills and attitudes.		

Personalized attention			
Methodologies	Description		
Seminar	Individual or small group tutoring.		
Problem solving			
Document analysis			

		Assessment	
Methodologies	Competencies /	Description	Qualification
	Results		



Guest lecture /	A1 A2 A9	Valorarase o traballo do alumnado, as súas respostas, o seu nivel de coñecemento,e	5
keynote speech		a súa participación activa no debate cos seus compañeiros.	
Seminar	A1 B2 B3 B5 B8	SESIÓN MAXISTRAL, SEMINARIOS, SOLUCIÓN DE PROBLEMAS: computaranse	20
		conxuntamente (45% da calificación global)	
Problem solving	A1 A2 A9 B2 B10 B11	SESIÓN MAXISTRAL, SEMINARIOS, SOLUCIÓN DE PROBLEMAS: computaranse	20
	C1 C4	conxuntamente (45% da calificación global)	
Objective test	A1 A2 A9 B2 B3 B5	Computará o 55% da calificación global.	55
	B8 B10 B11 C1		

Assessment comments

1. Assessment procedure. The assessment of this subject will be done

through a system whose sections and their respective weighting is detailed:

Assessment system (Weighting):

- Final examination (55 %)

- Continuous evaluation (45 %)

The continuous assessment (N1) will have a weight of 45% in the qualification of the subject and will be fundamentally telematic (Virtual Campus or Microsoft TEAMS). It will consist of presentations in the Virtual Campus of problems and practical cases (35%), in the evaluation of the student through questions and questionnaires during the course (5%) and in the oral presentation (works, reports, problems and practical cases) (5%). The final examination (N2) will have a weight of 55 % and will cover all the contents of the subject.

The final student?s score will be calculated by applying the following formula:

Final mark = 0.45 x N1 + 0.55 x N2

Being N1 the numerical mark corresponding to the continuous assessment (0-10 scale) and N2 the numerical mark of the final examination (0-10 scale). In any case, to pass the course, it is mandatory to achieve a minimum mark of 5.0 (0-10 scale).

2. Recommendations with regard to the evaluation.

The student should review the theoretical concepts introduced in the

various topics using the supporting material provided by teachers and the

literature recommended for each theme. The degree of success in the resolution

of the exercises provides a measure of the student's preparation to deal with

the final examination of the subject. Students who find difficulties in working

the proposed activities should consult with the teacher, with the goal that it

can analyze the problem and help solve these challenges.

3. Recommendations with regard to the recovery.

Teacher will discuss with students who do not successfully overcome the

evaluation process, and want it, the difficulties encountered in learning the

contents of the subject. The teacher will also provide additional material

(questions, exercises, exams, etc.) to reinforce the learning of the subject.

4. Others.

Attendance at face-to-face activities (face to face lectures, seminars and tutorials) is mandatory. The faults must be documentary supported, accepting reasons referred to in the University regulations.

	Sources of information
Basic	- P. Atkins, J. de Paula: Physical Chemistry, 10ª Edición; Oxford University- I. N. Levine: Principios de Fisicoquímica,
	6ª Edición; McGraw-Hill, 2014As ediciones previas ás especificadas tamén son válidas para esta materia A.R. West:
	"Solid State Chemistry and its Applications". Wiley, 2 ed., 2014 L.E. Smart, E.A. Moore: "Solid State Chemistry: An
	Introduction". CRC Press, 4 ed., 2012 R.Thomas : ?Practical Guide to ICP-MS?, CRC Press, Taylor & amp; Francis
	Group 2008- C.Stephan: ?Single-Particle ICP-MS Compendium? Perkin Elmer, 2016 - M.E.Schimpf, K.Cadwell,
	J.Calvin Giddings: ? Field-Flow fractionation handbook?, John Willey & amp; Sons, New York, 2000 - J.Janca :?
	Field-flow fractionation: analysis of macromolecules and particles?, Marcel Dekker, New York, 1988



Complementary	- A.I. Kirkland, S.J. Haigh: "Nanocharacterisation", 2ª Edición. RSC Publishing, 2015 S.R. Morrison: The Chemical
	Physics of Surfaces; 2nd ed.; Plenum Press, 1990 D. Myers: Surfaces, Interfaces and Colloids: Principles and
	Applications; VCH, 1999 S.E. Lyshevski (Editor): "Dekker Encyclopedia of nanoscience and nanotechnology" (7
	volumes), 3ª Edición. CRC Press, 2014 John P. Sibilia: ?A guide to materials characterization and chemical
	analysis?. VCH Publishers, 1998 C. Hammond: "The basics of Crystallography and Diffraction", 4ª Edición.
	International Union of Crystallography, Oxford University Press, 2015 C. Giacovazzo, editor ? Fundamentals of
	Crystallography? 3ª Edición. International Union of Crystallography, Oxford University Press, 2011 P.J. Goodhew:
	Electron Microscopy and Analysis. 3ª edición. Taylor & Francis, 2001 JP. Eberhart: "Structural and chemical
	analysis of materials : X-ray, electron and neutron diffraction, X-ray, electron and ion spectrometry, electron
	microscopy ". Wiley, 1991 Y. Leng: ?Materials Characterization. Introduction to Microscopic and Spectroscopic
	Methods?, 2ª Edición. Wiley-VCH, 2013 Ademais, recomendaranse para cada tema textos complementarios (artigos
	científicos, páxinas web, textos específicos) no momento de impartición da materia.

Recommendations
Subjects that it is recommended to have taken before
Subjects that are recommended to be taken simultaneously
/
/
Subjects that continue the syllabus
Other comments

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.