

		Teachin	g Guide		
	Identifying Data			2019/20	
Subject (*)	Applied Coordination Chemistry Code			610509110	
Study programme	Mestrado Universitario en Investig	gación Química	a e Química Industria	l (Plan 2017)	
		Descr	iptors		
Cycle	Period	Ye	ar	Туре	Credits
Official Master's Degree	e Yearly	Fir	rst	Optional	3
Language	Spanish				
Teaching method	Face-to-face				
Prerequisites					
Department	Química				
Coordinador	Fernandez Lopez, Alberto A.		E-mail	alberto.fernande	ez@udc.es
Lecturers	Fernandez Lopez, Alberto A.		E-mail	alberto.fernande	ez@udc.es
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Web					
General description	Este curso enmárcase na especia	alidade de Quír	nica Sintética dedica	do ao estudo da sín	tese e propiedades de diversas
	especies químicas. Dentro de dita	a especialidade	escolléronse algúns	tipos de compostos	s que, xa sexa polas súas
	especiais características, xa sexa	polas súas ap	licacións futuras, me	erecen un estudo de	tallado. Tendo isto en conta, a
	Química da Coordinación merece	un estudo apa	arte, tanto polas parti	culares característic	as dos compostos de
	coordinación como polos seus mé	étodos de sínte	ses. Desde outro pu	nto de vista, os char	mados complexos preséntanse
	nunha variedade estrutural enorme que vai desde especies de dimensión molecular pasando por agregados				
	supramoleculares, polímeros mor	no-, bi-, e tridim	ensionáis, ata chega	r ao chamados Met	al Organic Frameworks (MOFs)
	que forman redes tridimensionáis	ordenadas. Da	ada esta enorme vari	edade estrutural, no	on é de sorprender que o número
	de propiedades e aplicacións que	presentan sex	a tamén diverso. To	das estas razóns xu	stifican a súa inclusión tanto na
	especialidade de Química sintétic	a como nun Ma	áster dedicado ao es	tudo da Química.	

	Study programme competences / results
Code	Study programme competences / results
A1	Define concepts, principles, theories and specialized facts of different areas of chemistry.
A2	Suggest alternatives for solving complex chemical problems related to the different areas of chemistry.
A3	Innovate in the methods of synthesis and chemical analysis related to the different areas of chemistry
A8	Analyze and use the data obtained independently in complex laboratory experiments and relating them with the chemical, physical or
	biological appropriate techniques, including the use of primary literature sources
B1	Possess knowledge and understanding to provide a basis or opportunity for originality in developing and / or applying ideas, often within a
	research context
B2	Students should apply their knowledge and ability to solve problems in new or unfamiliar environments within broader (or multidisciplinary
	contexts related to their field of study.
B3	Students should be able to integrate knowledge and handle complexity, and formulate judgments based on information that was
	incomplete or limited, include reflecting on social and ethical responsibilities linked to the application of their knowledge and judgments.
B4	Students should be able to communicate their conclusions, and the knowledge and the reasons that support them to specialists and
	non-specialists in a clear and unambiguous manner
B7	Identify information from scientific literature by using appropriate channels and integrate such information to raise and contextualize a
	research topic
B10	Use of scientific terminology in English to explain the experimental results in the context of the chemical profession
B12	Being able to work in a team and adapt to multidisciplinary teams.
C1	CT1 - Elaborar, escribir e defender publicamente informes de carácter científico e técnico
C3	CT3 - Traballar con autonomía e eficiencia na práctica diaria da investigación ou da actividade profesional.
C4	CT4 - Apreciar o valor da calidade e mellora continua, actuando con rigor, responsabilidade e ética profesional.

Learning outcomes



Learning outcomes		Study program	
	cor	npetenc	es/
		results	
Students will be able to design new routes to prepare and isolate coordination compounds.	AC1	BC1	CC1
	AC2	BC2	CC3
	AC3	BC3	
	AC8	BC4	
		BC7	
		BC10	
tudents will be able to identify the chirality in mononuclear coordination complexes and identify its origin	AC1	BC1	CC4
	AC2	BC2	
	AC3	BC3	
	AC8	BC7	
		BC10	
Students will be able to describe the factors that imply activation small molecules after coordination to metal centres and their	AC1	BC1	
applications.	AC2	BC2	
	AC3	BC3	
	AC8	BC7	
		BC10	
		BC12	

	Contents
Торіс	Sub-topic
Structural properties of coordination compounds.	Structural properties in coordination compounds.
Characterization techniques	Characterization techniques.
	- Spectroscopic techniques
	- Spectrometric techniques
	- Diffraction techniques
	- Techniques based on magnetic properties
	- Other techniques
Activation of small molecules by coordination compounds	Structure and bonding in dioxygen and dinitrogen complexes: bonding modes.
	Synthesis of dioxygen and dinitrogen complexes.
	Structural characterization of dioxygen and dinitrogen complexes: NMR and IR
	spectroscopy, mass spectrometry, single crystal X-ray diffraction.
	Synthetic applications of dioxygen and dinitrogen complexes. Future perspectives.
Coordination compounds with applications in medicine:	Coordination compounds with applications in medicine: therapeutic and diagnose tools
therapeutic and diagnose tools	- Antitumor metalodrugs.
	- Types of complexes according to the metal centre and the structure
	- Therapeutic mechanism.
	- Synthetic methods.
	Application of metal complexes as radiopharmaceuticals.
	- Properties and characteristics of the ligands and the radioisotope
	- Diagnostic, (PET, SPEC) and therapeutic techniques.
	Contrast agents in magnetic resonance imaging (MRI)
	Proton relaxation agents T1 (Gd3+, Mn2+ y Fe3+) y T2. Parameters affecting the
	efficiency.
	- Agents based on saturation transfer by chemical exchange (CEST).
	- Agents based on other nuclei (, 31P).
	- Hyperpolarization.



ation compounds in the design of new materials:	Types of systems arising from the metal-ligand association. Isolated and extensive
tion polymers and MOFs. Properties and applications.	associations (coordination polymers and MOFs) Main structural characteristics and
	characterization.
	Synthetic strategies and general characterization techniques.
	Properties and applications.

	Planning	g		
Methodologies / tests	Competencies /	Teaching hours	Student?s personal	Total hours
	Results	(in-person & virtual)	work hours	
Seminar	A1 A2 A3 B1 B2 B3	7	21	28
	B4 B7 B10 B12 C1			
	C3 C4			
Mixed objective/subjective test	A1 A2 A3 A8 B1 B10	2	18	20
Guest lecture / keynote speech	A2 A3 B3 B7 B12 C4	12	13	25
Personalized attention		2	0	2

(*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

	Methodologies
Methodologies	Description
Seminar	Interactive sessions related to the different subjects with debates and exchange of opinions with students. Resolution of
	practical exercises (problems, test questions, interpretation and processing of information, evaluation of scientific publications,
	etc.)
Mixed	Tests designed to gauge the achievement of both theoretical and practical knowledge and the acquisition of skills and
objective/subjective	attitudes.
test	
Guest lecture /	The lectures in which the contents of the subject will be explained with the assistance of illustrative examples. The class slides
keynote speech	will be available, prior to the class.
	In some cases, if the number of students and their characteristics are adequate complementary methodologies as, for
	example, the case study or analyses of bibliographic sources might be used. The active participation of students will be
	encouraged.
	Attendance to lectures is not compulsory by highly advisable.

	Personalized attention
Methodologies	Description
Seminar	Individual or small group tutorials designed to support the students in the learning process. The student may attend to the
Mixed	tutorial freely to solve any doubts related to the subject. Tutorials are particularly important to solve the proposed problems or
objective/subjective	the assigned tasks.
test	Students granted the ?part time? condition will be given special attention through the tutorials which will be dedicated
	specifically to the solution of the problem sheets.

		Assessment	
Methodologies	Competencies /	Description	Qualification
	Results		
Guest lecture /	A2 A3 B3 B7 B12 C4	Continuous assessment of students through questions made during the classes.	10
keynote speech		Attendance and active participation will also be assessed	
Seminar	A1 A2 A3 B1 B2 B3	Resolution of problems and study cases. Attendance will also be assessed as well as	20
	B4 B7 B10 B12 C1	the active participation to de different activities.	
	C3 C4		



Mixed	A1 A2 A3 A8 B1 B10	Mixed test exam with questions and problems related to class contents.	70
objective/subjective			
test			

Assessment comments

The ?no presentado? mark will be granted to those students who do not attend to the mixed text.

Students

granted the ?part time student? condition will be assessed exclusively by the mixed text marks (100% of the subject mark)

	Sources of information
Basic	- J. Rivas Gispert (2000). Química de Coordinación. Omega
	- J. Rivas Gispert (2008). Coordination Chemistry . Weinheim: Willey-VCH
Complementary	

Recommendations
Subjects that it is recommended to have taken before
/
Advanced Structural Determination/610509103
Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus
Other comments
The student must know the basic principles of coordination
chemistry as, for example, the definition of coordination compound and
its components, as well as the bonding theories used to describe this type of
compounds

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.