		Teaching	Guide		
	Identifyir	ng Data			2020/21
Subject (*)	Physics 2 Code			610G01004	
Study programme	Grao en Química				
		Descrip	otors		
Cycle	Period	Yea	ır	Туре	Credits
Graduate	2nd four-month period	Firs	st	Basic training	6
Language	SpanishGalician				
Teaching method	Face-to-face				
Prerequisites					
Department	Física e Ciencias da Terra				
Coordinador	Rilo Siso, Esther		E-mail	esther.rilo.siso@	udc.es
Lecturers	Montero Rodríguez, María Belén		E-mail	belen.montero@	udc.es
	Rilo Siso, Esther			esther.rilo.siso@	udc.es
Web					
Contingency plan	subject that is the link between mathematics and chemistry in the sense of giving a formal formulation of scientific observations that establish laws and results without which you can not "close" the scientific method. The laws of physics provide the basic ingredients in which most sciences are supported, as well as instrumentation and measurement techniques used in all scientific fields, and especially in chemistry. Hence its importance and presence in the first year of the degree, since along with Physics 1 provides students with the necessary basis for understanding matters of other modules and courses for the degree.  1. Modifications to the contents				c method. The laws of physics ation and measurement nd presence in the first year of
	2. Methodologies  *Teaching methodologies that are  *Teaching methodologies that are  3. Mechanisms for personalized a  4. Modifications in the evaluation  *Evaluation observations:	e modified	ents		

	Study programme competences			
Code	Study programme competences			
A1	Ability to use chemistry terminology, nomenclature, conventions and units			
А3	Knowledge of characteristics of the different states of matter and theories used to describe them			
A12	Ability to relate macroscopic properties of matter to its microscopic structure			
A14	Ability to demonstrate knowledge and understanding of concepts, principles and theories in chemistry			
A15	Ability to recognise and analyse new problems and develop solution strategies			
A19	Ability to follow standard procedures and handle scientific equipment			
A20	Ability to interpret data resulting from laboratory observation and measurement			
A22	Ability to plan, design and develop projects and experiments			
A23	Critical standards of excellence in experimental technique and analysis			
A24	Ability to explain chemical processes and phenomena clearly and simply			

A25	Ability to recognise and analyse link between chemistry and other disciplines, and presence of chemical processes in everyday life
A27	Ability to teach chemistry and related subjects at different academic levels
B1	Learning to learn
B2	Effective problem solving
В3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
B5	Teamwork and collaboration
B7	Effective workplace communication
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life
C6	Ability to assess critically the knowledge, technology and information available for problem solving

Learning outcomes			
Learning outcomes	Study	y progra	amme
	COI	mpeten	ces
Have the minimum theoretical foundations that allow the understanding of the aspects of chemistry related to the electrical and	A1		C1
magnetic phenomena and vibratory motion and wave motion.	А3		
	A12		
	A14		
	A25		
Know how to reduce real problems to their most essential aspects and apply them to the field of chemistry		B1	C1
	A15	B2	СЗ
	A27	В3	C6
		B4	
		B5	
		В7	
Apply the basic laboratory techniques, including the necessary calculations and expressing the results appropriately. Use the	A19	B1	C3
material and apply the basic safety standards to work in a laboratory.	A20	B2	C6
	A22	В3	
	A23	B5	
	A24	В7	

Contents					
Topic Sub-topic					
1. Introduction to the study of the physic fields	1.1. Fields theory				
	1.2. Gravitational field				
2. Electricity	2.1. Electric field and potential.Capacity				
	2.2. Electric current and direct current circuits				
3. Magnetism	3.1. Magnetic field				
	3.2. Magnetic induction				
	3.3. Alternating current circuits				
4. Oscillations and waves	4.1. Oscillations				
	4.2. Waves motion				
	4.3. Electromagnetic waves				

Practical teaching: resistance measurement using a Wheatstone bridge, measurements of voltage, resistance and current in electrical circuits, light diffraction in a thread, simple pendulum, spring constant.

	Planning			
Methodologies / tests	Competencies	Ordinary class	Student?s personal	Total hours
		hours	work hours	
Guest lecture / keynote speech	A1 A3 A12 A14 A15	27	67.5	94.5
	A24 A25 A27 B1 B2			
	B3 C6			
Problem solving	A14 A15 A27 B1 B2	9	18	27
	B3 B4 B5 B7 C1 C3			
	C6			
Laboratory practice	A19 A20 A22 A23	15	0	15
	A24 B1 B2 B3 B5 C3			
	C6			
Mixed objective/subjective test	A1 A3 A12 A14 A15	2	0	2
	A24 A25 B2 B3 C6			
Personalized attention		1	0	1

	Methodologies
Methodologies	Description
Guest lecture /	During these sessions, teacher will explain lessons including different formats (theory, problems and general examples),
keynote speech	emphasizing the more important aspects and in the more difficult ones.
Problem solving	In this sessions, some problems related to theory contents explained before will be proposed and solved. Students must solve
	this problems and questions under teacher supervision, individually or in groups. There will be included in these classes
	activities that imply the participation of the pupils, that will contribute to the continuous assessment. So teacher can observe
	the difficulties of comprehension that every pupil presents in the resolution of problems.
Laboratory practice	Students will perform laboratory practice for the application of knowledge acquired in the keynote sessions and problem
	solving. With this methodology, they acquire skills needed to work properly in a physics lab, which includes the use of
	instruments for measurement, data processing and analysis of results of physic properties and magnitudes. A guide for each
	practice will be given to the student, and they will have all necessary material to mount and do them.
Mixed	It is the test for the evaluation of knowledge, which allows teacher assessing the level of student learning.
objective/subjective	
test	

	Personalized attention				
Methodologies	Description				
Laboratory practice	Students will be attended individually to help them to understand and resolve all problems related with the subject they can				
Problem solving	have. Moreover, teacher regularly invite students to tutorials with the intention of receiving the necessary guidance.				

		Assessment	
Methodologies	Competencies	Description	Qualification

Laboratory practice	A19 A20 A22 A23	Attendance to Laboratory practices is MANDATORY, so you cannot pass the course	15
	A24 B1 B2 B3 B5 C3	without making them. The highest mark that can be obtained is 1.5 points, and the	
	C6	minimum one required to pass them is 0.7. It will be evaluated on the basis of	
		participation and results delivery of each session, and a test that will take place during	
		the last session. Competences evaluated A19, A20, A22, A23, A24, B1, B3, B5, B7,	
		C3	
Problem solving	A14 A15 A27 B1 B2	Participation on the resolution of problems and exercises will be evaluated. Teacher	20
	B3 B4 B5 B7 C1 C3	may periodically collect exercises or questions proposed during these sessions.	
	C6	Competences evaluated: A1, A3, A12, A15, B1, B2, C1	
Mixed	A1 A3 A12 A14 A15	Final examination accounts for 35% of the final grade	65
objective/subjective	A24 A25 B2 B3 C6	During the term there will be partial exams whose maximum score will be 30% of the	
test		final grade. Competences evaluated: A1, A3, A12, A14, A15, B2, C1.	

## **Assessment comments**

## Exam mark should not

be less than 5 (up to 10). The final mark must

be 5 or higher to pass course, and will be calculated as follows: exam mark\*0.35+partial exams+laboratory+problem

solving. If a student, having a final mark higher than 5, fails

the minimum mark in any activity, he/she will have a mark of 4.5, i.e., Fail.

The evaluation of students in the second opportunity will follow the same criteria as at the first opportunity. The students tested in the second opportunity may only be eligible for honors if the maximum number of these for the corresponding course was not covered at the first opportunity. In the July opportunity will be saved the qualifications of Laboratory and Seminars of problems.

Students which due to justified reasons or for being enrolled part-time do

not participate in the ongoing evaluation activities volunteers, may do

equivalent work, consisting of delivery and explanation during sessions of individualized

tutoring bulletins problems and activities proposed in small group sessions.

The labs will be held according to the schedule published at the beginning

of the semester. The completion is mandatory, so it is necessary to overcome to pass the course.

For the rating of No Presented students they must not have participated in

activities totaling more than 25% of the final grade .

	Sources of information
Basic - Tippler & Osca (). Física para la ciencia y la tecnología . Reverté	
	- Sears, Zemansky, Young & Samp; Freedman (). Física Universitaria . Addison Wesley Longman
	- Fidalgo & amp; Fernández (). Física General. Everest
Complementary	- Burbano de Ercilla, Burbano García & Amp; Gracia Muñoz (). Problemas de Física. Mira
	- Lea & Burke (). Física, la naturaleza de las cosas. Paraninfo
	- Angel Franco García (2006). Física con ordenador. Curso interactivo de Física en internet.
	www.sc.ehu.es/sbweb/fisica/default.htm
	- (). Fisicalab. Plataforma de aprendizaje de física y matemáticas. www.fisicalab.com

	Recommendations
	Subjects that it is recommended to have taken before
Mathematics 1/610G01001	
Physics 1/610G01003	
	Subjects that are recommended to be taken simultaneously
Mathematics 2/610G01002	
	Subjects that continue the syllabus



Other comments

You need to have knowledge of physics and mathematics from high school.

(\*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.