		Teachin	g Guide			
	Identifying	Data			2019/20	
Subject (*)	Analytical Chemistry 2			Code	610G01012	
Study programme	Grao en Química				'	
		Descr	iptors			
Cycle	Period	Ye	ar	Туре	Credits	
Graduate	2nd four-month period	Sec	ond	Obligatory	6	
Language	Spanish					
Teaching method	Face-to-face					
Prerequisites						
Department	Química					
Coordinador	Turnes Carou, Maria Isabel E-mail isabel.turn			isabel.turnes@u	udc.es	
Lecturers	Moreda Piñeiro, Jorge		E-mail	jorge.moreda@	jorge.moreda@udc.es	
	Turnes Carou, Maria Isabel			isabel.turnes@u	udc.es	
Web						
General description	The aim of this subject is the study	of the sampli	ng strategies, sam	ple pre-treatment and	treatment methodologies and	
	separations techniques. Basic fundamentals of electroanalytical techniques (condutometry and potentiometry) are also					
	enclosed in the scope of this subject	ct				

	Study programme competences
Code	Study programme competences
A7	Knowledge and application of analytical methods
A15	Ability to recognise and analyse new problems and develop solution strategies
A16	Ability to source, assess and apply technical bibliographical information and data relating to chemistry
A17	Ability to work safely in a chemistry laboratory (handling of materials, disposal of waste)
A18	Risk management in relation to use of chemical substances and laboratory procedures
A19	Ability to follow standard procedures and handle scientific equipment
A20	Ability to interpret data resulting from laboratory observation and measurement
A21	Understanding of qualitative and quantitative aspects of chemical problems
A22	Ability to plan, design and develop projects and experiments
A23	Critical standards of excellence in experimental technique and analysis
A24	Ability to explain chemical processes and phenomena clearly and simply
A25	Ability to recognise and analyse link between chemistry and other disciplines, and presence of chemical processes in everyday life
B2	Effective problem solving
В3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
C1	Ability to express oneself accurately in the official languages of Galicia (oral and in written)
C6	Ability to assess critically the knowledge, technology and information available for problem solving
C8	Understanding role of research, innovation and technology in socio-economic and cultural development

Learning outcomes				
Learning outcomes	Learning outcomes Study program		amme	
	COI	mpeten	ces	
Knowledge and application of Sampling techniques and preparation of sample to the analysis.			C1	
	A16	В3	C6	
Application of main separation techniques	A7	B2	C1	
	A15	В3	C6	
	A21			

To acquire the basic skill in the laboratory of Analytical Chemistry	A17	B4	C1
	A18		C6
	A19		
	A20		
	A22		
	A23		
Ability to explain phenomena and processes related to Analytical Chemistry clearly	A7	В3	C1
	A24		C8
	A25		

	Contents
Topic	Sub-topic
Sampling techniques and preparation of sample	Sampling theory and methodology
	Sample pretreatament
	Sample preparation methods
Separation techniques	Precipitation
	Distillation
	Extraction
	ionic exchange
	Clasical electrophoresis
	Chromatography
Electroanalytical techniques	Conductometry
	Potentiometry
Laboratory experiments	Preparation of mussels samples for metal analysis
	Determination of fat content in food
	Concentration of trace elements in water by ion exchange
	Extraction of caffeine from beverages
	Separation of plant pigments by thin layer chromatography
	Conductimetric titrations: composition of strong and weak acids mixture. determination
	of acetylsalicylic acid in Aspirin

Planning			
Competencies	Ordinary class	Student?s personal	Total hours
	hours	work hours	
A7 A15 A17 A18 A19	20	20	40
A20 A22 A23 A24 B2			
B3 B4 C6 C8			
A7 A15 A20 A21 A22	8	20	28
A24 A25 B2 B4 C1			
A7 A16 A21 A22 A24	24	54.48	78.48
A25 B3			
A7 A15 A20 A24 B2	3	0	3
B3 B4 C1			
	0.52	0	0.52
	Competencies A7 A15 A17 A18 A19 A20 A22 A23 A24 B2 B3 B4 C6 C8 A7 A15 A20 A21 A22 A24 A25 B2 B4 C1 A7 A16 A21 A22 A24 A25 B3 A7 A15 A20 A24 B2	Competencies Ordinary class hours A7 A15 A17 A18 A19 20 A20 A22 A23 A24 B2 B3 B4 C6 C8 A7 A15 A20 A21 A22 8 A24 A25 B2 B4 C1 A7 A16 A21 A22 A24 A25 B3 A7 A15 A20 A24 B2 B3 B4 C1 B3 B4 C1	Competencies Ordinary class hours Student?s personal work hours A7 A15 A17 A18 A19 20 20 A20 A22 A23 A24 B2 B3 B4 C6 C8 20 A7 A15 A20 A21 A22 8 20 A24 A25 B2 B4 C1 A7 A16 A21 A22 A24 24 54.48 A25 B3 A7 A15 A20 A24 B2 3 0 B3 B4 C1 B3 B4 C1 0 0

Methodologies	
Methodologies	Description

Laboratory practice	In the laboratory sessions students will carry out the application of the theoretical concepts studied in the classroom. The
	scripts will have questions that the students will have to answer and deliver once finished the practices.
	During laboratory sessions, and on a simultaneous way to the realisation of the experiments, the student will have to
	elaborate a diary of laboratory that collect the calculations, the experimental procedures and the necessary settings. The
	professor will review the notebook of each student in each laboratory session.
Seminar	On site activities for small to very small groups in which the students must participate actively. Seminars will be mostly devoted
	to solve examples of real problems and numerical excercises. The problems are discussed and solved by the students
	following the guidance of the instructor.
Guest lecture /	The teacher will present the fundamental contents of each of the topics. For better learning, students will have to advance the
keynote speech	development of these sessions teaching materials suitable for your personal preparation.
Mixed	The test to evaluate the knowlege gained by the student will include both theoretical and numerical questions.
objective/subjective	
test	

	Personalized attention
Methodologies	Description
Seminar	The work developed by students in seminars and laboratory practice involves personal attention from the teacher both in the
Laboratory practice	resolution of questions as a guide to the preparation thereof, correction questionnaires, understanding fault indication etc.
	In addition, Professor mention to the student individually to discuss in more depth how their learning progress of matter.
	Moreover, all students can consult the teacher any aspect of the subject in the tutorial schedule established for this purpose
	Students being recognized officially as partial-time and entitled not to attend the lectures will be attended in a tutorships
	regime (set hour with teacher in advance).

Assessment				
Methodologies	Competencies	Description	Qualification	
Seminar	A7 A15 A20 A21 A22	It will assess the student's participation, the resolution of problems and numerical	10	
	A24 A25 B2 B4 C1	excercises, compliance dates for delivery or revision. Some short tests will be done		
		periodically in the seminar sessions and /or guest lecture to assess the evolution of		
		the student.		
Laboratory practice	A7 A15 A17 A18 A19	It will assess the performance of questions, abilities and skills of students in the	20	
	A20 A22 A23 A24 B2	experimental work, their ability to interpret the results, etc.		
	B3 B4 C6 C8			
Mixed	A7 A15 A20 A24 B2	It will assess the student's ability to express, summarize and develop theoretical	70	
objective/subjective	B3 B4 C1	aspects of the subject and the resolution of problems and numerical exercises.		
test				

Assessment comments

To pass the subject three basic requirements are needed:

- 1.-The laboratory sessions are mandatory.
- 2.-Reach a minimum qualification of 5 over 10 in the laboratory practice and in each mixed test.
- 3. Students who do not participate on the problem solving and do not carry out the short answer questions will score 0 in these sections (10%).
- -In the first and second time, students who not reach 5 points in practices evaluation, have the opportunity to, in addition to the mixed test, perform a specific test related to the labs. The score of this test especcífica replaced the grade obtained in practice for the overall rating.

The student will obtain the qualification of No Presented when the student does not assist to laboratory sessions and to the final mixed exam.

In the context of "continuous evaluation" the "second opportunity of July" is a second opportunity of realisation of the mixed test. Therefore, the laboratory practice, and seminars, will keep the qualifications obtained along the course, whereas the qualification of the mixed test of second opportunity will substitute to the one obtained in the mixed test of the first opportunity.

The students evaluated in the "second opportunity" only will be able to opt to matrícula of honour if the maximum number of these for the corresponding course has not covered in its whole in the "first opportunity".

- For students with recognition of dedication and part-time academic exemption waiver assistance, conducting laboratory practices are mandatory and it will be provided within the flexibility to allow coordinating schedules and material and human resources. They shall be deemed exempt from the keynote sessions while assistance will be provided to the greatest number of seminars. Therefore, these students will be evaluated by the grades obtained in laboratory practices (20%), in the mixed test (70%) and in the activities of the seminars (10%). If they can not attend the seminars will make a mentored work.

	Sources of information
Basic	- CÁMARA, C.; FERNÁNDEZ, P.; MARTÍN-ESTEBAN, A; PÉRZ-CONDE, C.; MIQUEL VIDAL (2002). Toma y
	Tratamiento de Muestra. Madrid, Ed. Síntesis
	- SKOOG, D.; WEST, D.N.; HOLLER, F.J.; CROUCH, S.R. (2005). Fundamentos de Química Analítica . Madrid, Ed.
	Thomsom, 8 ^a edición
	- HARRIS DANIEL C (2007). Análisis Químico Cuantitativo. Barcelona, Ed. Reverté, 3ª Edición
Complementary	- VALCARCEL, M. & Dímez, A. (1988). Técnicas Analíticas de Separación. Barcelona, Ed. Reverté
	- HARVEY, D. (2002). Química Analítica Moderna. Madrid, Ed. McGraw-Hii
	- GAVIRA VALLEJO, J.M. HERNANZ GISMERO, A. (2007). Técnicas Físicoquímicas en Medio Ambiente. Madrid,
	Ed.Libreria UNED

Subjects that it is recommended to have taken before	
General Chemistry 3/610G01009	
Analytical Chemistry 1/610G01011	
Subjects that are recommended to be taken simultaneously	
Chemistry Laboratory 2/610G01032	
Subjects that continue the syllabus	
Instrumental Analytical Chemistry 1/610G01013	
Instrumental Analytical Chemistry 2/610G01014	
Other comments	

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.