		Teachin	g Guide		
	Identifying	Data			2020/21
Subject (*)	Chemistry, Information and Society			610G01031	
Study programme	Grao en Química				
		Descr	riptors		
Cycle	Period	Ye	ear	Туре	Credits
Graduate	1st four-month period	Sec	ond	Obligatory	6
Language	Spanish				
Teaching method	Hybrid				
Prerequisites					
Department	Química				
Coordinador	Penedo Blanco, Francisco Jose		E-mail	francisco.penedo	o.blanco@udc.es
Lecturers	Penedo Blanco, Francisco Jose		E-mail	francisco.penedo	o.blanco@udc.es
Web					
General description	In this subject the main aspects rela	ated to the So	cience's developn	nent, the scientific inform	ation sources, and the Science
	Society and Industry relationships a	are addressed	d. The key object	ves to achieve are the b	uilding of a critical and ethical
	view of scientific work, as well as th	ne compreher	nsion of the every	time strong relationship	between Science and Society.
Contingency plan	1. Modifications to the contents				
	- There will be no variations				
	2. Methodologies				
	*Teaching methodologies that are n	maintained			
	- None				
	*Teaching methodologies that are n	modified			
	- All methodologies will go to onli	ine mode.			
	3. Mechanisms for personalized att	ention to stud	dents		
	- Email: Daily. It will be used to m			ual meetings and to solv	e doubts.
	- Moodle: According to the needs			Ü	
	- Teams: Sessions adapted to the			according to the schedu	lo sot in the class calendar
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	4. Modifications in the evaluation - There will be no modifications *Evaluation observations:	he teaching g			le set ill the class calendar.

	Study programme competences
Code	Study programme competences
A16	Ability to source, assess and apply technical bibliographical information and data relating to chemistry
A18	Risk management in relation to use of chemical substances and laboratory procedures
A21	Understanding of qualitative and quantitative aspects of chemical problems
A23	Critical standards of excellence in experimental technique and analysis
A24	Ability to explain chemical processes and phenomena clearly and simply
A25	Ability to recognise and analyse link between chemistry and other disciplines, and presence of chemical processes in everyday life
A28	Acquisition, assessment and application of basic principles of industrial activity, organisation and task management

B2	Effective problem solving
В3	Application of logical, critical, creative thinking
B4	Working independently on own initiative
B5	Teamwork and collaboration
B6	Ethical, responsible, civic-minded professionalism
B7	Effective workplace communication
C3	Ability to use basic information and communications technology (ICT) tools for professional purposes and learning throughout life
C4	Self-development as an open, educated, critical, engaged, democratic, socially responsible citizen, equipped to analyse reality, diagnose
	problems, and formulate and implement informed solutions for the common good
C5	Understanding importance of entrepreneurship, and knowledge of resources available for people with business ideas
C6	Ability to assess critically the knowledge, technology and information available for problem solving
C7	Acceptance as a professional and as a citizen of importance of lifelong learning
C8	Understanding role of research, innovation and technology in socio-economic and cultural development

Learning outcomes			
Learning outcomes		progra	
To know the different ways to obtain information and to communicate research results throughout history and today.	A16	В3	C6
	A24	В7	
	A25		
	A28		
To know the methods of current and past research, and the social and individual environment influences.	A16		C6
	A23		C7
	A25		
To learn the ways to obtain written, audiovisual and online information in Chemistry.	A16	B2	СЗ
	A24	B4	
	A25		
	A28		
To know and understand the pathways leading to the results in the process of chemical research.	A16	В3	C8
	A25	B5	
	A28	В7	
To know, learn and critically evaluate the research ethics. To know and judge the responsible behaviour. To observe and	A18	В3	C4
correct mistakes, bad practices and negligence in daily work.	A21	В6	C8
	A23	В7	
	A25		
	A28		
To understand the past and present relationship between Society, Science and Industry, and its strong social influences.	A24	B2	C4
	A25	В3	C5
	A28	В6	C8
		В7	

	Contents
Topic	Sub-topic
SECTION I: Origin and development of research and theories	Topic 1. "This is we are" - An Overview of the History of Science and
	Chemistry
	Topic 2. Scientific vs Chemical Revolutions. Why did they evolve at a different pace?
	Topic 3. Methods and Practice in Science

SECTION II: Communication of results	Topic 4. Information Sources
	Topic 5. Decisions, publication and evaluation of results. Chemical Societies
	Topic 6. Responsible Science
	Topic 7. Intellectual property and Patents
	Topic 8. Science Popularization
SECTION III: Risks and Benefits of Chemistry and Chemical	Topic 9. Evolution of Science-Technology interaction.
Industry	Topic 10. The Chemical Industry. Will it always be an environmental problem?
	Topic 11. Scientists and Military Industry, a controversial marriage.

	Planning			
Methodologies / tests	Competencies	Ordinary class	Student?s personal	Total hours
		hours	work hours	
Guest lecture / keynote speech	A16 A18 A21 A25 B6	30	30	60
	C4 C7 C8			
Supervised projects	A16 A18 A21 A23	10	40	50
	A24 A28 B2 B3 B5 B7			
	C5 C6			
ICT practicals	A16 B2 B3 B4 C3 C6	4	8	12
Practical test:	A16 B2 B3 C3	2	6	8
Mixed objective/subjective test	A16 A21 A24 A25 B3	3	15	18
	B6 C4 C5 C6 C7 C8			
Personalized attention		2	0	2

	Methodologies
Methodologies	Description
Guest lecture /	The teacher presents and explains the fundamental concepts of each topic.
keynote speech	
Supervised projects	At small group sessions the student will solve the problems proposed by the teacher, who will supervise their work. Case
	studies linking Chemistry and mass media, health, food, environment, etc., will be discussed. The students must individually
	submit a final report or perform an oral statement of their work.
ICT practicals	At small group sessions in the Computer classroom, the student will practice the search for specific information in Chemistry,
	using online bibliographic databases.
Practical test:	A practical exam of searching online bibliographic databases, to evaluate the knowledge acquired in the ICT practicals
Mixed	A final exam containing multiple choice, short answer and essay questions. Its objective is to assess the knowledge acquired
objective/subjective	by the students, as well as their critical thinking and their ability to reason, synthetize and create texts.
test	

	Personalized attention
Methodologies	Description



Supervised projects

Throughout all sessions of the small group classes, the students' tutoring will be enhanced, helping them to ask and solve doubts.

Mandatory personal attention (2 hours) will be scheduled throughout the classes' period via interactive tests, which intend to assess the student's progress to the learning aims of the subject.

The student can request individual tutoring at the teacher's office, on the schedule published on the Faculty website.

In the specific case of part-time students or exemption of assistance, seminars and supervised work will be led through interactive online support, email or moodle, with a similar schedule to that of small group classes.

		Assessment	
Methodologies	Competencies	Description	
Mixed	A16 A21 A24 A25 B3	This exam includes multiple choice, short answer and essay questions.	
objective/subjective	B6 C4 C5 C6 C7 C8		
test			
Supervised projects	A16 A18 A21 A23	Evaluation is carried out taking into account the following aspects:	34
	A24 A28 B2 B3 B5 B7	- Active participation and critical thinking showed throughout the debates in the	
	C5 C6	classroom.	
		- Synthesis and reasoning skills reflected in the works presented.	
Practical test:	A16 B2 B3 C3	The evaluation will take into account the accuracy and skill in finding information in	33
		bibliographic databases, with a series of exercises to be performed in the Computer	
		classroom.	

Assessment comments

To pass the course there are two basic requirements:

- 1) Regular attendance to all evaluable activities; the attendance is compulsory, except for students with part-time or waiver of assistance, as indicated below
- 2) To achieve a minimum score of 4.5 (out of 10) in every evaluable activities. If this minimum score is not achieved in any of these activities, the final mark will be Fail (4.0). To pass the subject, the overall rating may not be less than 5 (out of 10).

"Not attended" assessment mark will be applied when the conducted activities add up to less than 33% of the total score.

Only the marks obtained in Seminar and Supervised projects may be retained

for the second opportunity, provided they exceed the minimum of 4.5

previously quoted. Marks falling below implies its corresponding assessment must be repeated. The

specific retake schedule of these two tests depends on

the number of students who have to attend them, and will be published at the end of the

first assesment.

The mixed test's mark obtained in the second opportunity will replace the first one's.

Students assessed at the second opportunity may only qualify for Honors mark if

the maximum number of licenses for the course were not exhausted on the

first one.

- No mark will be retained for subsequent courses, i.e. the teaching-learning process including assessment, will start over, which means that the students must complete all scheduled activities.

Related to the part-time students: For all purposes they are considered as students with standard registration, except they requested waiver of assistance as well, in such case the following paragraph shall apply.

Related to the students with exemption from assistance: the attendance to the mixed test is compulsory, but all the other assessed activities can be conducted online, including the practices of databases search, and ranked as if they were in attendance, so the design of a specific protocol it is not necessary.

	Sources of information
Basic	- P. J. Bowler, I.R. Morus (2007). Panorama general de la ciencia moderna. Editorial Crítica, Madrid
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	- David C. Lindberg. (2002). Los inicios de la ciencia occidental . Editorial Paidós, Barcelona
	- W.H. Brock (1998). Historia de la química, serie: Ciencia y Tecnología . Editorial Alianza Editorial, 1998, Madrid
	- A.F. Chalmers (1993). ¿Qué es cosa llamada ciencia?. Siglo XXI, Madrid
	- Patricia Fara (2009). Breve historia de la ciencia . Editorial Ariel, Barcelona
	Páxinas web de utilidade:- http://www.udc.es/biblioteca- http://echa.europa.eu/- http://ec.europa.eu/index_es.htm-
	http://www.epo.org
Complementary	

Recommendations
Subjects that it is recommended to have taken before
Subjects that are recommended to be taken simultaneously
Subjects that continue the syllabus
Other comments



- For non English-speaking students: IT IS RECOMMENDED TO HAVE A MEDIUM OR ADVANCED LEVEL OF COMPREHENSION OF ENGLISH TEXTS.- Writing skills using common computer tools are needed. Additionally, it's very important to have at least an intermediate skill level using a portable slideshow application, such as Microsoft PowerPoint or OpenOffice Impress.

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.