		Teaching G	iuide			
	Identifyir	ng Data			2015/16	
Subject (*)	Física			Code	610G02002	
Study programme	Grao en Bioloxía				'	
	<u>'</u>	Descripto	ors			
Cycle	Period	Year		Туре	Credits	
Graduate	2nd four-month period	First		FB	6	
Language	SpanishGalicianEnglish					
Teaching method	Face-to-face	Face-to-face				
Prerequisites						
Department	Física					
Coordinador	Domínguez Pérez, Montserrat		E-mail	montserrat.dominguez.perez@udc.es		
Lecturers	Cabeza Gras, Oscar		E-mail	oscar.cabeza@udc.es		
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Web						
General description	Physics subject try to teach the b	pasic concepts of pl	hysics and its a	applicability to Biology.	Those concepts are necessary	
	understand many natural phenomena that will be studied in other fields and subjects of the Biology Grade.					

	Study programme competences		
Code	Study programme competences		
A22	Describir, analizar, avaliar e planificar o medio físico.		
A26	Deseñar experimentos, obter información e interpretar os resultados.		
A30	Manexar adecuadamente instrumentación científica.		
A31	Desenvolverse con seguridade nun laboratorio.		
B1	Aprender a aprender.		
B2	Resolver problemas de forma efectiva.		
В3	Aplicar un pensamento crítico, lóxico e creativo.		
B4	B4 Traballar de forma autónoma con iniciativa.		
B5	B5 Traballar en colaboración.		
B8	Sintetizar a información.		
B10	Exercer a crítica científica.		

Learning outcomes			
Learning outcomes		Study programme	
	CO	mpeten	ces
To know the basic physical concepts in the different parts of Physics, as: Mechanics, Fluids, Waves, Thermodynamics,	A22	B2	
Electromagnetism and Optics.			
Know how to relate the physical concepts with the biology phenomena.	A26	B10	
Apply the theoretical knowledge to the resolution of basic physical problems, mainly focused to resolve biologycal phenomena.		B1	
	A26	B2	
		B8	
To know and to use the methodologies, bibliographic sources and technical concepts corresponding to Physics, using the	A30	В3	
scientific method to its study.		B4	
Learn the basic Physics Laboratory techniques, like to measure fundamental physical magnitudes as density, viscosity,	A26	B5	
surface tension, specific heat	A30	B8	
	A31		

	Contents
Topic	Sub-topic
Introducction to Physicas	Physical Magnitudes.
	Measuremente, dimensions and unities.
Vector Analysis	Vectors. Types. Components
	Operations with vectors
	Momentum of a vector
Motion Descripcion	Kinematics. Movement. Characteristics
	Speed and acceleration
	Types of movements.
Motion and Forces	Dynamics. Newton Movement Laws
	Movement Quantity
	Gravity Force
	Types of forces
	Friction
Equilibrium Study	Static Principles
	Center of mass
	Moment of inertia. Steiner Theorem
Biomecanics. Scale Laws	Muscular strength. Momentum
	Scale Laws. Metabolic Rate
Mecanical Energy. Conservation	Work and Power
	Kinetic and Potential Energy
	Energy Conservation
Deformed Media	Elasticity. Hooke's Law
	Traction. Young's Module
	Lateral Contraction. Poisson Coefficient
	Compresibility Coefficient
	Flexion
	Cutting
	Torsion
Ideal Fluids. Statics and Dynamics	Density
	Pressure. Magnitudes, unities and measurement
	Fundamental Equation of Hydrostatics
	Pascal and Archimedes Principles
	Continuity Equation
	Bernouilli's Theorem. Aplications
Real Fluids	Viscosity
	Fluids Flow modes
	Reynolds' Number
	Laminar Regime. Poiseuille Equation
	Viscosity Measurement. Ostwald Viscometer
	Movement of solids through fluids
Surface Phenomena	Molecular Forces. Surface Tension
	Laplace's Law
	Capillarity. Jurin's Law

Harmonical and Wavy Movements	Simple Harmonic Movement. Pendulum
	Wave Types
	Wavy Movement Equation
	Speed of wave propagation
	Energy and intensity of the wavy movement
	Doppler Effect
Acoustics. Ultrasounds	Speed of Sound
	Noise Quality
	Sound Sensation
	Reverberation
	Ultrasounds
Thermodynamics. Temperature.	Thermodynamical Systems
Thombodynamics. Temperature.	Thermodynamical variables
	Thermodynamical variables Thermodynamical processes
	Zero Principle of Thermodynamics. Temperature.
	Temperature Measurement. Escales and thermometers
Gas Study.State Equations	Ideales Gas. Laws
ous stady. State Equations	State Equation
	Real Gas.Van der Waals' Equation
	Kinetic Theory of Gas
First Principle of Thermodynamics	Heat and Work.
First Finiciple of Thermodynamics	
	Internal Energy
	Thermodynamic Work
	P-V Diagram
	Nature and Effects of Heat
	Heat Transmission
	Internal Energy
	First Principle of Thermodynamics
	Entalpía
	Transformaciones de los gases ideales
Cocond Dringiple of Thermodynamics	Thermal Machine Concept
Second Principle of Thermodynamics	·
	Two forms for the Second Principle of Thermodynamics
	Carnot Cicle
	Entropy Concept. Entropy Calculation
Concepts on electricity and bio-magnetism	Electrical Charge. Coulomb's Law
	Electrical Field and Potential
	Dipoles
	Capacity. Capacitors
	Current Intensity. Ohm's Law
	Electrical resistivity and conductivity
	Electrical Current Energy
	Magnetic Forces
	Leyes de Laplace y Faraday
	Corrientes alternas

Radiation and radioactivity	De Broglie's relationship	
	Bonding Energy. Mass Loss	
	Fision and fusion	
	Radiactivity. Atom Splitting	
	Physical and Biological Dosimetry	
	Biological Effects of Radiation	
Notions on Optics	Electromagnetic waves	
	Lens and Mirrors	
	Optical Instruments	

	Planning	l		
Methodologies / tests	Competencies	Ordinary class	Student?s personal	Total hours
		hours	work hours	
Introductory activities	B1	1	0	1
Document analysis	A26 B8	0	3	3
Laboratory practice	A26 A30 A31 B5 B8	14	14	28
Problem solving	A22 A26 B1 B2 B8	8	24	32
Objective test	A22 A26 B2 B10	4	0	4
Guest lecture / keynote speech	A22 B1 B3 B10	28	42	70
Supervised projects	B3 B4 B5 B8 B10	0	9	9
Personalized attention		3	0	3

	Methodologies
Methodologies	Description
Introductory activities	The first day of the course we will give to each student the program of this Subject, the metodology we will follow, the
	evaluation criteria, and also a detailled calendar with all activities.
Document analysis	We will inform to students the necessary bibliographical data, both for problems, theory and assisted jobs. Thus, they could
	revise and increase the aspects explained in the classroom. The individual tutorials will help also in those aspects.
Laboratory practice	Along the six Laboratory sessions students will work in couples, doing five complete practices. A guide for each practice will be
	given to teh student, and they will have all necessary material to mount and do them. All time students will be assisted by its
	teacher to resolve all doubts and help if necessary.
	At the end of practice time, each couple will present a memory including the job performed and the obtained results.
	Prior to the Laboratory sessions there will be a room session to explain the basis of experimental uncertainties and graphical
	representations.
Problem solving	After the theoretical exposition of each lesson, there will be Seminars (with a reduced number of students) to resolve problems
	to apply the theory studied. The proposed problems for each lesson will be given to the students before each of those
	sessions as bulletins. There we will include the numerical solution of each problem, to allow students evaluate themselves
	after doing them individually. Those bulletins will be of two different types: some of them General (the same for all students of
	the three groups), and other complementary bulletins specific for each reduced group. Not all problems will be completely
	resolved in the
	Seminars, but only those more difficult.
Objective test	There will be two written exams about the theory and numerical problems saw in classroom. The first one at the middle of the
	course and the second one at the end. The students that pass each of those exams will have that part of the subject passed
	for the Final exams of June (and Jully).
Guest lecture /	The basic content of the different parts of the Subject will be explained by the teacher in this sessions, trying to involve
keynote speech	students in the learning process. At the end of each session will be in the Moodle the material used that day to facilitate pupils
	its study.
Supervised projects	Voluntarily the students can do complementary work. That will be do in pairs of students and will be focused in applications of
	Physics to Biology, including notions of electricity, optics and modern physics.



Personalized attention		
Methodologies	Description	
Document analysis	Students will be attended individually to help them to understand and resolve all problems related with the subject they can	
Laboratory practice	have, including: bibliography, problems of the bulletin, the complementary work In resume all doubts they can have in the	
Supervised projects	study and comprenhesion of physic subject.	

		Assessment	
Methodologies	Competencies	Description	Qualification
Laboratory practice	A26 A30 A31 B5 B8	The total calification of Laboratory will be the 15 % of the final calification. This will	15
		have three different parts:	
		- A 5% will correspond with the Laboratory note book given to the teacher with the five	
		practices made.	
		- Other 5% will represent the evaluation of the practice made the sixth day of	
		Laboratory.	
		- The last 5% will come from the evaluation of a test exam that all students must do	
		with the official exams in June or July.	
		The attendance to the session previous to Laboratory is mandatory to be evaluated.	
		The laboratory calification will be pass if you obtain a minimum of 0.7 pts (on 1.5 pts).	
Problem solving	A22 A26 B1 B2 B8	The participation in the Seminars will represent a 5% of the final calification.	5
Objective test	A22 A26 B2 B10	The theoretical exams made along the course will count a 21 % to the final calification,	70
		while the problems exam will be a 49 % of that.	
		The addition of both califications (theory and problems) must be 4/10 points minimum	
		to pass the subject.	
Supervised projects	B3 B4 B5 B8 B10	The voluntary job calification will count a 10% of the global one.	10

Assessment comments

The NP (non presented) qualification will be given to those students that do not attend all Laboratory sessions, and they have not attend to the final tests. Also, if you have only Lab qualification the note would be Fail (no NP). In the July opportunity will be saved the qualifications of Laboratory, Voluntary job and Seminars of problems.

Remember that Laboratory job qualification (including if it were less than 0.7/1.5) is mandatory to pass the subject Laboratory assistance is mandatory (exceptions will only be those given in Artigo 12 of the "Normas da avaliación, revisión e reclamación

das cualificacións dos estudos de Grao e Mestrado Universitario"). The laboratory practices must be done in the official calendar published at the beginning of the term.

If a student, having an average qualification higher than 5, fails the minimum qualification in any activity, he/she will have a qualification of 4.5, i.e., Fail.

Students with a partial matriculation have to do the laboratory practices and those activities related to pass the subject, being the 15% of global qualification. They can do the voluntary job (10%), and so the Ojective test will count 75%.

Sources of information		
Basic	- Cussó, López y Villar (2004). Física de los procesos biológicos. Barcelona. Ariel	
	- Kane y Sternheim (1994). Física. Barcelona. Reverté.	
	- Jou, Llebot y Pérez (1994). Física para las ciencias de la vida . Barcelona. Mc. Graw- Hill	



Complementary	- Tippler, P (2005). Fisica I y II. Barcelona. Reverté
	- Ortuño (1996). Física para biología, medicina, veterinaria y farmacia . Barcelona. Crítica
	- Burbano y Burbano (1991). Problemas de Física . Barcelona. Mira
	- Feynman, R. P. (2005). The Feynman lectures on physics. Vol. I, II and III. Addison-Wesley
	- Serway, R.A. and Jewitt, J.W. (2014). Physics for Scientist and Engineers. USA. Cengage Learning
	- Young, H.D. and Geller, R.M. (2007). Sears and Zemansky's College Physics. USA. Pearson
	- Wilson, J.D. and Hernández-Hall, C.A. (2015). Physics Laboratory Experiments. USA. Cengage Learning

Recommendations	
Subjects that it is recommended to have taken before	
Matemáticas/610G02003	
Subjects that are recommended to be taken simultaneously	
Subjects that continue the syllabus	
Other comments	

(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.