



| Teaching Guide | | | | |
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| Identifying Data | | | 2020/21 | |
| Subject (*) | Physico-chemistry and quality of water | | Code | 632844203 |
| Study programme | Mestrado Universitario en Enxeñaría da Auga (plan 2012) | | | |
| Descriptors | | | | |
| Cycle | Period | Year | Type | Credits |
| Official Master's Degree | 1st four-month period | First | Obligatory | 6 |
| Language | English | | | |
| Teaching method | Face-to-face | | | |
| Prerequisites | | | | |
| Department | Enxeñaría Civil | | | |
| Coordinador | Delgado Martin, Jordi | E-mail | jorge.delgado@udc.es | |
| Lecturers | Barrientos Rodríguez, Victor | E-mail | victor.barrientos@udc.es | |
| | Delgado Martin, Jordi | | jorge.delgado@udc.es | |
| | Vázquez González, Ana María | | ana.maria.vazquez@udc.es | |
| Web | caminos.udc.es/hosting/masteragua/ | | | |
| General description | This subject is aimed at presenting some basic concepts about the physics and chemistry of natural waters as well as some key ideas about water quality. Elemental water chemistry concepts are combined with other practical topics like natural water sampling, data analysis and graphical representation. More advanced contents include the description of processes governing the variability of the chemical composition of natural waters in its different reservoirs (precipitation, continental lotic and lentic systems, ground water, sea water). | | | |
| Contingency plan | <p>1. Modifications to the contents</p> <p>There are no changes in contents or its scope</p> <p>2. Methodologies</p> <p>The preferred class method will be based on student attendance to classes. In case mobility becomes limited, classes will proceed with the support of online methodologies (e.g. Teams)</p> <p>*Teaching methodologies that are maintained</p> <p>All the teaching methodologies are maintained.</p> <p>*Teaching methodologies that are modified</p> <p>In case of limitations in mobility, lab practices will become desk practices. Field trips will transform into guided audiovisulas (tematic documentaries) supported by explanations and common discussions.</p> <p>3. Mechanisms for personalized attention to students</p> <p>Personalized attention will be maintained as usual but taking advantage of telematic tools (e.g. Teams) in case mobility becomes limited</p> <p>4. Modifications in the evaluation</p> <p>The assessment of the subjet will be performed in the same basis as non-contingent situation (including weighting). In the caso of short answer tests, they will be performed with the aid of telematic tools (e.g. Moodle)</p> <p>5. Modifications to the bibliography or webgraphy</p> <p>No modifications apply</p> | | | |

| Study programme competences | |
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| Code | Study programme competences |
| A1 | Knowledge, understanding and capacity to apply legislation related with water engineering during professional development. Capacity to analyse the working mechanism of the economy and public and private management of water |



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| A2 | Capacity to resolve basic physical problems of water engineering and theoretic and practical Knowledge of the chemistry, physics, mechanics and technologic properties of the water |
| A5 | Knowledge of the basic concepts about ecology applied to water engineering. Capacity to act in the respectful way and enriching way about the environment contribution to the sustainable development. Capacity to analyse the ecological quality of water. Knowledge of the basic principles of the ecology and basic understanding of the working continental water systems |
| A16 | Knowledge of the chemical basis of water which totally condition its behaviour in nature and its uses. Understanding and knowledge of the different water regulations for quality at local, national and European level |
| A19 | Knowledge of advanced water treatment with different conclusions: depuration, re-use, purification, elimination of nutrients and regeneration treatments |
| A20 | Use and management of measuring equipment in the field and in the laboratory. Knowledge of the methodology of control process and the determination of design parameters for water treatment processes |
| A21 | Knowledge of water quality control models. Capacity to analyse and propose solutions to problems in water quality control |
| A25 | Knowledge and understanding of water in different situations: the working of ecosystems, environmental factors with the purpose of to make an inventory of medium, applying the methodology to value the impact and its use in studies and evaluations of the environmental impact. |
| B1 | To resolve problems effectively |
| B2 | To apply critical thinking, logic and creativity |
| B3 | To work individually with initiative |
| B4 | To communicate effectively in work surroundings |
| B5 | Continuous recycling of knowledge in a general perspective in a global situation of water engineering |
| B6 | Understanding of the need to analyse history to understand the present |
| B7 | Facility to integrate in multidiscipline teams |
| B8 | Capacity to organize and plan |
| B9 | Capacity for analysis, synthesis and structure of information and ideas |
| C1 | To understand the importance of the enterprising culture and to know the means at the reach of the enterprising people |
| C2 | To value knowledge critically, technology and available information to resolve problems that they will face |
| C3 | To assume as a professional and citizen the importance of learning throughout life |
| C4 | To value the importance of the investigation, innovation and technology development in the social ?economic advance and cultural in society |
| C5 | To posses and understand knowledge that gives a base or oportunity to be original in the development and for applications of ideas, often in the context of investigation |
| C6 | The students must be able to apply the acquired knowledge and their capacity to resolve problems in new surrandings or not well known within wider contexts (or multidiscipline) related with the study area |
| C7 | The students must be able to integrate knowledge and to affront the complexity to formulate judgements from information that, been incomplete or limited, include reflexions about social responsibilities and ethics related to the application of the knowledge and judgments |
| C8 | The students must be able to comunicate their conclusions, knowledge and the last reasons that support them, to spezialated publics and not spezialated in a clear and unambiguous way. |
| C9 | The student must possess the learning ability with permits them to continues to study in a manner wich will be in a great measure self directed and individual |

Learning outcomes

| Learning outcomes | Study programme competences |
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|---|------|-----|-----|
| Learning the basic principles of water chemistry. | AC1 | BC1 | CC1 |
| | AC2 | BC4 | CC2 |
| | AC5 | BC5 | CC3 |
| | AC16 | BC6 | CC4 |
| | AC19 | BC9 | CC5 |
| | AC20 | | CC6 |
| | AC21 | | CC7 |
| | AC25 | | CC8 |
| | | | CC9 |
| Learning the basic principles of the analytical techniques aimed at quantifying the concentrations of water contaminants and their constituents. | AC2 | BC1 | CC2 |
| | AC16 | BC2 | CC3 |
| | | BC4 | CC4 |
| | | BC5 | |
| | | BC7 | |
| | | BC9 | |
| Ability to plan and execute sampling surveys for water chemistry | AC1 | BC1 | CC4 |
| | AC2 | BC2 | |
| | AC20 | BC3 | |
| | AC21 | BC5 | |
| | AC25 | BC7 | |
| | | BC8 | |
| | | BC9 | |
| Ability to establish relationships between physico-chemical data and the chemical state of a water body or the prescribed legal environmental quality objectives. | AC1 | BC2 | CC2 |
| | AC25 | BC5 | CC3 |
| | | BC7 | CC4 |
| Ability to perform statistical descriptions relative to the chemical quality of water. | AC2 | BC1 | CC2 |
| | AC16 | BC2 | CC3 |
| | AC20 | BC4 | CC4 |
| | AC21 | BC7 | |
| | | BC8 | |
| | | BC9 | |
| Ability to perform graphical representations of water chemistry | AC2 | BC1 | CC2 |
| | AC25 | BC2 | CC3 |
| | | BC3 | CC4 |
| | | BC8 | |
| | | BC9 | |
| Learning basic hydrochemical processes | AC16 | BC1 | CC3 |
| | AC19 | BC2 | CC4 |
| | | BC7 | |
| | | BC9 | |
| Learning the basic principles of hydrochemical modelling | AC21 | BC1 | CC4 |
| | | BC2 | |
| | | BC7 | |
| | | BC9 | |

| Contents | |
|----------|-----------|
| Topic | Sub-topic |



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| Basics of water chemistry | <p>Structure and properties of water</p> <ul style="list-style-type: none">- Phase diagram of water- Density, salinity, heat capacity, viscosity- Oceanic stratification and thermohaline circulation- Phase transformations of water- Stereochemistry of the water molecule <p>Basic chemistry concepts</p> <ul style="list-style-type: none">- Ponderal laws- Mass conservation- Mol and stoichiometry- Concentration units- Intensity and capacity properties <p>Colligative properties</p> <ul style="list-style-type: none">- Adhesión, cohesion and capillarity <p>Chemical bonding and aqueous interactions</p> <ul style="list-style-type: none">- Types of chemical bonding- Aqueous interactions- Emulsions and solutions <p>Chemical equilibrium and solubility</p> <ul style="list-style-type: none">- Thermodynamic systems and laws- Components, phases and species- Collision theory and chemical reactions- Mass action law and the equilibrium constant- Le Chatelier's Principle- Chemical kinetics and reaction rates |
| Sampling and monitoring | <p>Planning a water quality survey</p> <p>Routine and special analyses</p> <p>Water sampling: Tools and methodology</p> <p>Sample pre-treatment and preservation</p> <p>In situ versus laboratory parameter determination</p> <p>Sampling water systems</p> <ul style="list-style-type: none">- Ground water sampling and special equipment- Precipitation- Surface water (streams and rivers)- Lakes and reservoir sampling |



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| Basic analytical techniques and quality assessment of water analysis | <p>Experimental measurements</p> <p>Basic statistics</p> <ul style="list-style-type: none">- Statistical moments- Distribution functions and non-parametric statistics- Quantiles- Outliers <p>Basic analytical chemistry:</p> <ul style="list-style-type: none">- Precision- Accuracy- Error and bias- Calibration and analytical limits <p>Quality assessment:</p> <ul style="list-style-type: none">- Recommendations and rules-of-thumb <p>Quantitative and qualitative analyses</p> <p>Selection of instrumental analytical techniques:</p> <ul style="list-style-type: none">- Titrimetry- Spectrometric methods- Chromatographic methods |
| Graphical analyses of water quality data | <p>Basic graphical assessment</p> <ul style="list-style-type: none">- Single water samples- Multiple water samples <p>Advanced plotting and analysis techniques</p> <ul style="list-style-type: none">- Correlations and false correlations- Complex relationships- Time- and flow-adjusted concentrations- Time trend analyses- Time series analyses <p>Analyses tools:</p> <ul style="list-style-type: none">- Time trends- PAST |
| Interpretation of the quality of natural waters (Part I) | <p>The water cycle and the global energy budget</p> <p>Precipitation</p> <ul style="list-style-type: none">- Components of precipitation (dry, bulk, wet, hail, fog, etc.)- Precipitation sampling- Rain/forest/soil interactions- Smog and photochemical smog- Meteorological drivers and rain shadows- Chemical composition of precipitation- Sea spray- Acid rain- Global effects on precipitation- Critical loads- Local effects in precipitation |
| Interpretation of the quality of natural waters (Part II) | <p>Rivers and Streams</p> <ul style="list-style-type: none">- Basins and watersheds- River processes- Hyporrheic and riparian zones- Diel cycles- Major constituent origin and processes- Space and time dependencies in riverine systems |



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| Interpretation of the quality of natural waters (Part III) | <p>Lakes and Reservoirs</p> <ul style="list-style-type: none"> - Fresh water environments and ecological zoning - Lake types - The Aral Sea disaster - Special cases: Reservoirs, pit lakes and subglacial lakes - Residence time - Morphometrical studies: Methodology and descriptors - Energy budgets in lakes and reservoirs - Thermal classification of lakes and reservoirs - Light, attenuation and transparency - Oxygen - The cycles of C, N and P and their coupled systems |
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| Planning | | | | |
|---|--|----------------------|-------------------------------|-------------|
| Methodologies / tests | Competencies | Ordinary class hours | Student's personal work hours | Total hours |
| Guest lecture / keynote speech | A1 A2 A5 A16 A19 A21 A25 B5 | 30 | 30 | 60 |
| Seminar | A1 A2 A5 A16 A19 A20 A21 A25 B1 B2 B3 B4 B5 B6 B7 B8 B9 C1 C2 C3 C4 C5 C6 C7 C8 C9 | 30 | 30 | 60 |
| Personalized attention | | 30 | 0 | 30 |
| (*)The information in the planning table is for guidance only and does not take into account the heterogeneity of the students. | | | | |

| Methodologies | |
|--------------------------------|---|
| Methodologies | Description |
| Guest lecture / keynote speech | Regular lectures where the main theoretical contents of the subjects are regarded |
| Seminar | <p>Field trips and laboratory practice</p> <p>Field trips will be organized so that the student can put into practice a part of the knowledge acquired in the subject</p> <p>The students will go to the laboratory where they will put into practice the knowledge acquired to:</p> <ul style="list-style-type: none"> -Make the design of a field survey -To carry out the necessary analysis to obtain the value of the different physical-chemical parameters of the water samples collected in the organized field campaigns <p>Prior to the implementation of the work in the laboratory, the student will perform a basic theoretical preparation for each proposed practice, which will consist of reading the script to know the objective of the practice, know what he will do and why, know perfectly the management of the equipment that will be used and perform the necessary calculations for its experimental development. Before starting the practical session, the student will be called to assess if he is ready to start the practice</p> |

| Personalized attention | |
|--------------------------------|---|
| Methodologies | Description |
| Seminar | Personalized attention to be provided for the seminars and tutorings. These will require an adequate planning in order to make compatible the availability of teachers and students. These sessions may be telematic (e.g. Teams) in case that presentality becomes limited |
| Guest lecture / keynote speech | |



| Assessment | | | |
|--------------------------------|--|--|---------------|
| Methodologies | Competencies | Description | Qualification |
| Seminar | A1 A2 A5 A16 A19 A20 A21 A25 B1 B2 B3 B4 B5 B6 B7 B8 B9 C1 C2 C3 C4 C5 C6 C7 C8 C9 | The attendance to the seminars and the work being developed at the seminars will be considered for the final mark | 50 |
| Guest lecture / keynote speech | A1 A2 A5 A16 A19 A21 A25 B5 | The knowledge of the concepts developed at the magistral lectures will be assessed and considered for the final mark | 50 |

| Assessment comments |
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| <p>-Tests: Short answer and exercises.</p> <p>The evaluation of the theoretical part of the units of the subject will be done through a test-type examination at the end of the semester. The development of the teaching material will led to the resolution of practical problems that will be assessed at the end of each topic through a focus-control. The weighting over the final mark of this partial assessment will be 50%.-Field trips and visits. Field trips will be organized so that the student can put into practice part of the knowledge acquired in the subject -Laboratory practices.</p> <p>The students will develop in the laboratory some of the knowledges acquired in order to: - Plan a water sampling survey - To carry out the necessary analyses to obtain the value of the different physical-chemical parameters of the water samples collected in their planned field surveys Prior to the implementation of the work in the laboratory, the student will receive a basic or contextual theoretical background for each proposed practice, which will consist of: a) reading the script to know the objective of the practice; b) know what to do and why; c) understand best laboratory management procedures in order to make good and safe use of the equipment. Before starting the practical session, the student will be called to assess if he is ready to start the practice.</p> <p>At the end of the course, students will present a personal work related to field work and laboratory work whose partial weight over the total mark of the subject will be 50%</p> |

| Sources of information | |
|------------------------|---|
| Basic | <ul style="list-style-type: none">- James I. Drever (1997). The Geochemistry of Natural Waters: Surface and Groundwater Environments (3rd Edition). Prentice Hall- Werner Stumm and James J. Morgan (1996). Aquatic Chemistry: Chemical Equilibria and Rates in Natural Waters (3rd Ed.). Wiley Interscience- C.A.J. Appelo and D. Postma (2005). Geochemistry, Groundwater And Pollution (2nd Ed.). Balkema- John D. Hem (1985). Study And Interpretation of the Chemical Characteristics of Natural Water. U.S. Geological Survey- Arthur Hounslow (1995). Water Quality Data: . Lewis Publishers |
| Complementary | |

| Recommendations |
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| Subjects that it is recommended to have taken before |
| Subjects that are recommended to be taken simultaneously |
| Subjects that continue the syllabus |
| Other comments |
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(*)The teaching guide is the document in which the URV publishes the information about all its courses. It is a public document and cannot be modified. Only in exceptional cases can it be revised by the competent agent or duly revised so that it is in line with current legislation.